

harmful consequences on the living organisms of these ecosystems [4-5]. The consequences of severe metal poisoning depend on the physical chemical factors of water, but also on the amount and duration of the action. Among aquatic organisms, fish are highly sensitive to pollution and as such are quite suitable as bioindicators for water quality monitoring [6].

Research into different types of fish has shown that heavy metals can lead to changes in the physiological activity and biochemical parameters of various tissues of these organisms [2, 7-9]. Heavy metals pose a risk to aquatic ecosystems due to their bioaccumulation. They can be accumulated and biomagnified in water, sediment and the aquatic food chain [10-11]. Since fish are a good source of protein for humans, their poisoning with heavy metals can also endanger human health [12-13]. Studies have shown that antioxidative vitamins such as C and E, which are widespread in many food products, have been shown to have a mitigation effect in heavy metals toxicity [14-16]. These two vitamins have been shown to have chelating abilities, affecting the reduction in the amount of accumulated metals in different tissues to various organisms, including fish [17-18]. The aim of the current work is to investigate in parallel the sensitivity of various tissues to the accumulation of heavy metals, and the chelating effect of tocopherol (vitamin E) and ascorbic acid (vitamin C) either alone or jointly in accumulated heavy metals (Pb, Cd and Hg) in different tissues (liver, gills, muscles and plasma) of common carp (*Cyprinus carpio*) treated with heavy metals.

Material and Methods

Common carp (*Cyprinus carpio*) was obtained from a small fish pound in the village of Janjevo (central Kosovo) near Prishtina. Their average weight was 150 ± 5 g and average length 20 ± 1 cm. Gender differences in the current research were not taken into consideration. Fish were placed in glass aquariums filled with tap water and acclimatized under laboratory conditions for three weeks. Physicochemical conditions of water during the acclimatization period and experiment were: pH 7.5 ± 0.5 and temperature $17\pm 2^\circ\text{C}$. The fish were fed once a day with commercial food while the water was changed every 48 hours. The fish involved in the experiment after the acclimatization period were divided into control and experimental groups. The experimental group (n24) for seven days was exposed to a heavy metal mixture containing lead acetate ($\text{Pb}(\text{CH}_3\text{COO})_2$) 0.3 mg/l, cadmium chloride (CdCl_2) 0.4 mg/l and mercuric chloride (HgCl_2) 0.001 mg/l. During exposure time, water in aquaria was changed every 24 hours. After heavy metals exposure, six fish were sacrificed and from them we took liver, muscle, gills and blood for heavy metal analysis. The remaining group (n18) was divided into three other groups that have undergone dietary supplements with vitamins for two weeks. Since vitamin C is water-soluble, it has been dissolved with

aquarium water in the amount of 5 mg/l. On the other hand, vitamin E is supplemented with food because it is fat-soluble. Food with vitamin E content was prepared in the laboratory. To the 500 g blended commercial food pellets we added 55 ml of soy oil and 500 mg of vitamin E. The pellets formed from such food mixture were dried in the oven for several hours at 100°C . One group of fish (n6) as supplement received only vitamin C, the other group (n6) only vitamin E, while the third group (n6) received vitamins C and E jointly. After two weeks, the fish of the three groups were sacrificed and the liver, gills, muscle and blood were taken for analysis of heavy metals content.

Heavy metals content (Cd, Pb, Ni and Cu) was determined by atomic absorption spectrophotometry (AAS). Statistical processing of results was done using a one-way analysis of variance (ANOVA) method including least significant difference (LSD) post-hoc test.

Results

The concentrations of heavy metals (Pb, Cd and Hg) in different tissues (liver, muscle, gills and plasma) of the carp (*Cyprinus carpio*) are presented in Tables 1-4 in terms of mean value and standard deviation (\pm), including LSD values. The control group is compared with the groups treated with heavy metals and the groups treated with heavy metals are compared with the vitamin-treated groups.

Based on results presented in all tables it is noted that the amount of Pb is higher in comparison to Cd and Hg and that gills are tissues where Pb is recorded in the largest quantity (Table 1) compared to other tissues (Tables 2-4). The highest level of cadmium is recorded in the liver of the control group and that of exposed to heavy metals compared to other tissues (Table 1). In liver, exposure to heavy metals caused a significant increase of lead and cadmium accumulation compared to the control group. From results presented in Table 1, it is seen that mercury is an exception because the level of accumulation was increased, but not significantly. Compared to the group exposed to metals, the level of accumulated metals in fish that have received vitamin supplements (C, E and C + E) significantly decreased. In terms of lead and mercury, vitamins C and C+ E jointly have demonstrated a better chelating effect compared to vitamin E separately. Regarding the level of mercury, vitamin C has lowered it below detection level. Meanwhile, with regard to Cd, the best chelating effects have been shown by vitamins C and E separately than these two vitamins jointly. The order of accumulated metals in liver of all groups is: lead>cadmium>mercury.

The results presented in Table 2 show that the highest level of accumulation of lead and mercury has been recorded in gills of all groups of fish and the order of accumulated metals was as well as on the liver (Pb>Cd>Hg). Unlike liver, in gills exposure to heavy metals (Pb, Cd and Hg) caused a significant

research. Vitamin E and C jointly create protection against lipid per oxidation. Vitamin C is able to recycle oxidized vitamin E, recovering the antioxidant ability of vitamin E [32, 34]. Ebuehi et al. (2012) indicate that oral administration of vitamins C and E significantly reduced the blood lead concentration, ameliorates the hepatic damage and significantly reduced the oxidative stress in the brain of rats [35]. A study by Rendón-Ramírez et al. (2014) with employees exposed to Pb (73 µg Pb / dL blood), also argued the synergistic effect of vitamins C and E. A year after oral vitamin C and E supplementation (1 g daily vitamin C and 400 IU daily vitamin E), lipid peroxidation in erythrocytes was reduced to between 47.1% and 69.4%, which were no longer statistically different from those of non-Pb exposed workers. The total antioxidant capacity in erythrocytes also returned to values between 58.9% and 67.7% in workers exposed to Pb after treatment, a level similar to those in exposed non-Pb workers [36].

Results of the present work indicate that the intake of vitamins (C, E and C + E) has significantly decreased the amount of accumulated heavy metals in all investigated tissues.

Conclusions

The observations of our data regarding the accumulation of heavy metals showed that gills and liver are sites of accumulation of heavy metals compared with muscles and blood (gills>liver>muscles>plasma), and the order of metal level in all investigated tissues is Pb>Cd>Hg.

Antioxidants (vitamins C, E and C + E), either separately or jointly, have been shown to have a good chelating effect on reducing the level of accumulated heavy metals in tissues involved in the research. However, their chelating ability was different.

From this point of view, it can be seen that the use of vitamins C and E separately has been shown to have a similar effect on reducing the amount of heavy metals accumulated in the tissues involved in the research. However, based on the results we obtained, it can be concluded that supplementation of vitamins (C + E) jointly is much more effective than in case these vitamins are separately applied. This may be due to the synergistic effect that produces the combination of vitamins C and E together.

Conflict of Interest

The authors declare no conflict of interest.

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