

Table 1. Organic C and N contents of soil, C sources and cow dung liquid.

	Organic C	Ammonium	Nitrate	Total Nitrogen	C to N ratio
Soil (g/kg)	4.72	0.02	0.003	0.44	11.75
Corn cob (g/kg)	923.5	0.27	0.11	8.53	55.51
Woodchips (g/kg)	801.7	0.14	0.36	1.86	264.89
Wheat straw (g/kg)	794.1	0.75	0.38	12.36	55.36

Note: Moisture content of soil was 12.29%; corn cob, woodchips and wheat straw were dried.

gases, can remove nitrate from aquatic ecosystems irreversibly [15-18]. Nitrate and available carbon source concentrations are key controlling factors of heterotrophic denitrification. The availability of carbon source becomes the limitation of denitrification while nitrate reaches a relatively high level, which may happen in agricultural drainage settings [19, 20].

A number of studies focused on the selection and optimization of carbon sources. Liquid carbon sources, such as methanol, ethanol and methanoic acid, could sustain relatively high denitrification rates. But for the chance of secondary pollution and costs, liquid carbon sources haven't been widely used to reducing nitrate in surface and subsurface water [21-25]. Woody materials with low cost and high permeability can support steady but relatively low denitrification rates [26-30]. Wheat straw, rice hulk and newspaper can sustain higher denitrification rates, but they degraded much more rapidly, which may not sustain long-term denitrification [31-34]. A number of polymer materials were also used for denitrification, but more studies are needed to lower their costs [35-38]. To select carbon substrates reaching a balance among availability, cost, denitrification rate and longevity will be significant to reduce nitrate in surface and subsurface water.

Studies on soil amendments had found that co-composting of cattle manure with organic waste can efficiently enhance lignocellulose degradation [39-41]. The role of cow dung liquid was evaluated in acidification of lignocelluloses materials, which indicated that lignocellulose decomposing microorganisms contained in cow dung liquid may stimulate the acidification and increase dissolved organic concentration in the system [42]. Following these evidence, this study evaluated the influences of cow dung liquid with different carbon sources to stimulate denitrification in laboratory.

Materials and Methods

Three different carbon sources used in this study were: corn cob, wood chips (predominately *Populus* spp.) and wheat straw. These carbon sources were ground to 1 to 2 mm size to avoid the effect of particle size on rates of nitrate removal. The C and N contents of these materials are listed in Table 1.

The influence of cow dung liquid with different carbon sources on denitrification was determined in anaerobic incubations where 5 g carbon material (dry weight basis) was mixed with 5 g of field moist subsoil and certain amount of cow dung liquid (0, 10 or 20 mL) in 250-mL glass jars. All jars received 1 mL of a nitrate solution that contained 20 mg $\text{NO}_3\text{-N}$ mL^{-1} and distilled water to keep the liquid volume at about 200 mL. The mixture of field moist subsoil, nitrate solution and distilled water resembles conditions in agricultural drainage water.

The cow dung used was collected in Dangyang City, China. 200 g fresh cow dung was fully mixed with distilled water and filtrated by 0.5 mm strainer, and the dissolved part was diluted to 1000 mL. The soil used to inoculate the microcosms was subsoil taken 50 cm below the surface of loamy sand soil located in a cotton field in Wuhan, China and the C and N contents were listed in Table 1. Jars prepared as described above without any carbon source (subsoil only) served as the contrast. Jars were incubated for 90 days at 24.5 to 25.5°C in an anaerobic growth chamber with a N_2 gas atmosphere. A sufficient number of jars were prepared so that three replications could be destructively sampled on 6 dates spaced 15 days apart. Initial substances added in different treatments was listed in Table 2.

Throughout the experiments $\text{NO}_3\text{-N}$ and $\text{NH}_4\text{-N}$ concentrations as well as chemical oxygen demand (COD) in the jars were monitored every 2 or 3 days and re-treated with certain amount of the nitrate solution to sustain nitrate concentrations between 50 to 200 mg/L. Jars were shaken for 30 min before a sample was taken and after the $\text{NO}_3\text{-N}$ solution was added. At 15 days intervals, three replicates of each treatment were sacrificed for accurately aqueous-phase analysis of $\text{NO}_3\text{-N}$, $\text{NH}_4\text{-N}$ and COD, as well as solid-phase analysis of weight. Contents in jars were mixed thoroughly and poured into 500 mL high-density polyethylene bottles, then centrifuged for 30 min at 6000 rpm.

The supernatant was decanted, filtered, and used for analysis. The remaining soil and solid carbon substrates were rinsed two more times to remove remaining $\text{NO}_3\text{-N}$ and $\text{NH}_4\text{-N}$ in the moist pellet by adding 100 mL distilled water, mixing thoroughly, and centrifuging as described above. These rinse water

Table 5. Mass reduction of solid remains in 90 days (unit: g).

Days	CC-0	CC-10	CC-20	W-0	W-10	W-20	WS-0	WS-10	WS-20
15	1.841	1.800	2.019	0.477	0.256	0.570	1.927	2.126	2.229
30	2.011	2.115	2.182	0.510	0.489	0.612	2.015	2.228	2.338
45	2.208	2.421	2.345	0.511	0.514	0.641	2.221	2.287	2.394
60	2.270	2.425	2.537	0.556	0.519	0.699	2.246	2.369	2.532
75	2.350	2.568	2.638	0.611	0.535	0.705	2.357	2.428	2.762
90	2.563	2.654	2.735	0.656	0.563	0.760	2.428	2.596	3.018

the availability of carbon sources in our study. COD had also been chosen as a key factor in studies on denitrification kinetics [47, 48]. The analysis of COD at 15-d intervals of each treatment has been shown in Fig. 5.

Initial COD of each incubation system was high and COD decreased sharply in the first 15 days. COD decreased steadily and slowly from the 15th day to the 90th day. COD of all treatments from highest to lowest was: wheat straw, corn cob and woodchips. Treatments with cow dung liquid added had higher COD than those using the same carbon substrates without cow dung liquid. 20 mL of cow dung liquid can result in a higher COD level than 10 mL. As one controlling factor of denitrification, COD can influence nitrate removal rates. The correlation analysis of COD and average nitrate removal rates of each sampling was listed in Table 4 (99% confidence level).

In woodchips treatments, there was an obvious correlation between COD and nitrate removal rates. But for corn cob and wheat straw treatments, no obvious correlation can be found. The differentials were determined by the three carbon substrates themselves: Corn cob and wheat straw contain some polysaccharose and protein, which degraded rapidly and lead to a high COD level. Woodchips degraded much more slowly owing to its stable structure, so low COD levels existed in woodchips treatments. Relatively lower COD could limit the nitrate removal rates, as a result, an obvious correlation between COD and nitrate removal rates existed in woodchips treatments.

Mass Decrease of Remaining Solid

Analysis of the mass decrease of remaining rinsed solid materials showed a certain relationship with cumulative $\text{NO}_3\text{-N}$ removed and COD (Table 5). On the 15th day, the mass decrease proportions of corn cob and wheat straw treatments were 35% to 45%, which revealed that easily degradable component of corn cob and wheat straw weighed at least 30% of the total mass. Masses of woodchips treatments decreased 5% to 11% proportioned to initial mass. Greater mass decrease of carbon substrates resulted in a higher COD level in relevant treatments. On the 90th day, 51%-55% of corn cob, 48%-60% wheat straw and

11%-15% woodchips have degraded. The decrease proportions of corn cob and wheat straw were close to the organic carbon contents of these two materials, which resulted in the decline of nitrate removal rates in the later period of incubation. The degrading rates of woodchips were low, for which woodchips may sustain long-term denitrification.

Conclusions

This study chose corn cob, woodchips and wheat straw as carbon sources for denitrification. The mixture of field moist subsoil, nitrate solution and distilled water in glass jars resembled conditions in agricultural drainage water. The study included the stimulation of carbon sources and the influences of cow dung liquid added. The results indicate:

1) Corn cobs, wood chips and wheat straw can stimulate denitrification and sustain continuous denitrification for 90 days.

2) In the woodchips treatment, $\text{NO}_3\text{-N}$ removal was carried on at lower rates than the other two carbon sources. Corn cob and wheat straw contain some polysaccharose and protein, which were more easily utilized by denitrifiers, while woodchips degraded slowly.

3) Mixed with cow dung liquid, further denitrification was stimulated in all three carbon treatments. In woodchips treatments, adding 20 mL cow dung liquid remove more $\text{NO}_3\text{-N}$ removed than 10 mL. In corn cob and wheat straw treatments, the addition of cow dung liquid got more $\text{NO}_3\text{-N}$ removed when carbon sources were adequate.

4) Monitoring of ammonia nitrogen and organic nitrogen of each treatment indicated that both the concentrations stayed at a relatively low level and denitrification was the dominant mechanism of $\text{NO}_3\text{-N}$ removal.

5) In woodchips treatments, correlation analysis indicated an obvious correlation between COD and nitrate removal rates. Analysis of the mass decrease of remaining rinsed solid materials revealed a greater mass decrease of carbon substrates resulted in higher COD levels in relevant treatments.

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