Short Communication

# Assessing Toxic Elemental Concentrations in Marine Fish *Trachurus capensis* (Cape Horse Mackerel) and Implications for Public Health

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#### Abstract

While fish is considered a healthy component of the human diet, consumption of fish with high levels of trace metals in their flesh constitutes a public health risk as trace metals have been proven to be toxic. We investigated the concentrations of toxic elements in seawater and also in different body parts of the fish *Trachurus capensis* caught near Durban, South Africa, using inductively coupled plasma-atomic emission spectrometry (ICP-AES). The highest metal concentration in fish body parts was observed for Pb, followed by Zn. Significantly higher levels of Mn were observed in fish gills as compared to the tissue (muscle) and fish frame. With respect to bioaccumulation, significantly higher Pb levels were observed in fish tissues compared to As, Cr, and Mn. In the frame, significantly higher Pb levels were observed compared to all other metals except Ba. There were no significant differences in the concentrations of different metals in fish gills. Overall, the toxic metal concentrations in the muscle of cape horse mackerel were below levels of concern for human consumption as defined by the FAO and WHO.

Keywords: Trachurus capensis, public health, toxic metals, bioaccumulation

## Introduction

Contrary to the historical belief that the ocean is immune to pollution due to its vastness, anthropogenic activity has made oceans susceptible to pollution [1]. Commercial wastes from industries are considered the major potential source of pollutants finding their way into the sea – particularly heavy metals and petroleum hydrocarbons [2]. Heavy metals can bioaccumulate to toxic concentrations in aquatic food webs, resulting in serious environmental and ecological consequences [3]. In the aquatic food-web, fish stand at the higher levels and may accumulate high concentrations of some toxic metals from both seawater and the food they eat [4-5].

Fish, however, is considered a healthy food in the human diet because of its high nutritional benefits related to proteins of high biological quality, attractive lipid composition, beneficial minerals, and useful vitamins [6]. Due to the lipid solubility of heavy metals and their resistance to several degenerative processes in animal tissue, different organs of fish can bioaccumulate trace metals to considerably higher concentrations than those

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found in the water around them [7]. Accumulated toxic metals from fish could easily interact with human genetic materials and other bio-macromolecules, leading to organ distortion, mutation, nervous disorder, kidney malfunctioning, bone diseases, and cancer [8]. Arsenic, cadmium, chromium, copper, lead, nickel, and zinc are the most commonly found toxic trace elements in nature, which enter water sources by natural means and through human activities such as wastewater discharge [9] and are the most likely to accumulate in fish organs. Consequently, determining heavy metal accumulation in commercial fish is an important task in order to calculate the viable risk to human health due to fish consumption [10].

*Trachurus capensis* is a pelagic fish species commonly called mossbunker or cape horse mackerel that forms a large population in the surface layers of the ocean (pelagic zone). They are common along the entire coast of South Africa and Namibia, with the highest population between the west coast and the Agulhas Bank. It is a common "by-catch" species that is a high source of protein and is exported to rural South Africa and other African countries. While bioaccumulation of toxic elements in edible fish tissues remains a health concern, relatively few studies have investigated the levels of metals in some fish species from the Durban coast [11-12]. The main aim of this work, therefore, was to estimate the concentration of toxic trace elements (Al, As, Pb, Cr, Mn, Zn, and Cu) in seawater and the bait fish T. capensis collected from selected popular fishing spots off the Durban coast, since they are an important component of the human diet in this area.

#### **Materials and Methods**

Three sites were chosen in and around the harbour of Durban, South Africa:

- Harbour mouth (S 29°52.017' E 31°03.961') immediately outside the harbour
- Container (S 29°49.478' E 31°04.535'), situated about
  5 km north of the harbour mouth
- Cuttings (S 29°58.404' E 30°59.287'), situated about 10 km south of the harbour where the southern outfall waste pipes are connected to the ocean (Fig. 1)

*Trachurus capensis* were caught by rod and reel. Immediately after fishing, all the fish were washed with seawater and then with fresh tap water, placed on ice, and brought to the laboratory on the same day for further analysis.

The fish were dissected into three sections, namely tissue, frame, and gills. The samples were then dried at 105°C for 8 h, preceded by ash drying at 650°C for 6 h. Triplicate 0.5 g ash samples were then digested with diacid (30 ml of HCL and 10 ml of HNO<sub>3</sub> in 3:1 ratio) on a hot plate maintained at 100°C until all materials were dissolved completely. Samples were then cooled at room temperature and transferred into 25 ml volumetric flasks containing 2.5 ml of indium (internal standard) for metal



Fig. 1. Map of Durban Coast in South Africa (Sampling sites are marked) Source: CSIR, South Africa.

analysis. The samples were subjected to analysis for various trace metals using inductively coupled plasmaatomic emission spectrometry (ICP-AES).

The bioaccumulation factor (BAF) in this study was predicted using the equations of Vaseem and Banerjee [13]. The EDI was calculated by factoring in the average metal concentration in a whole fish (this study), the average daily consumption of fish by an adult person [14], the average weight of an adult African [15], and the permissible tolerable daily intake of heavy metals [16].

The statistical package for social sciences (IBM SPSS Statistics 23) was used for data analysis. Analysis of variance (ANOVA) was used to determine statistically significant differences in the concentrations of metals in water and in fish body parts at a 0.05 level of significance using the least significant difference (LSD) as the post hoc test.

### **Results and Discussion**

The concentration of metals in seawater averaged 0.3 mg/L for all metals and in all study sites. The only slight but statistically insignificant deviation (P>0.05) was noted for Zn, whose concentration in seawater was 0.38 mg/L at the site container. The average metal concentration in different fish body parts is presented in Fig. 2. This average combines values for fish caught in each of the sampling sites off the Durban coast. The highest metal concentration in all body parts assessed was observed for Pb, followed by Zn, while the lowest metal concentration in any body part was observed for

Mn. Compared to other studies that evaluated metal concentrations in mackerel, higher metal concentrations were observed in our study than in previous studies (Table 1). Reported results in literature showed that metal accumulation in fish muscle varied widely depending on location [17], season [18], behavior, and feeding habits [19].

When the selective tendency of fish body parts to bioaccumulate metals was analyzed, significantly higher levels of Mn were observed in fish gills as compared to the tissues (muscle) (P = 0.005) and fish frame (P = 0.008). The differences in the concentration of the rest of the metals in various fish body parts were not significant (P>0.05).

The average concentration of metals in fish tissue (muscle) in this study was far greater than the average recorded by Elnabris et al. [14] in a range of fish collected from different points of the Gaza Strip. Metals have also been shown to have varying affinities for adsorption to fish tissues mainly based on varying uptake, deposition, and excretion rates [20]. When the bioaccumulation of various metals with respect to fish body parts were compared, significantly higher Pb levels were observed in fish tissue compared to As (P = 0.048), Cr (P = 0.05), and Mn (P = 0.044). The differences in the concentrations of other metals in fish tissue were not significant. In the fish frame, significantly higher Pb levels were observed compared to all other metals (P<0.05) except Ba (P>0.05). Other significant metal concentration differences were between As and Zn (P = 0.03), Cr and Zn (P = 0.027), Cu and Zn (P = 0.037), and Mn and Zn (P = 0.025). There were no significant differences in metal concentration in the gills. Overall, there was significantly higher Pb concentration in all fish body parts compared to all other metals (P = 0.003), with the exception of Zn (P = 0.185). The calculated bioaccumulation factor for each metal confirmed these statistical assertions.

The overall average metal bioaccumulation in this study was found to be in the order of Pb > Zn > Ba > Al > Cu > As > Cr > Mn. These findings contrast with the findings of Elnabris et al. [14], who reported higher concentrations of Zn, Ni, Cu, and Mn as compared to Pb, and the findings of Alam et al. [21], who reported higher concentrations of Cu, Cr, and Zn compared to Pb.

Fish are an important part of a healthy diet because of their richness in omega-3 fatty acids, essential minerals, and vitamins [22]. However, because of their trophic level, fish normally bioaccumulate toxic metals – especially pelagic fish [23] – from their food, sediments, and water, as shown by the results of this study. In this regard, the negative health effects of heavy metals in fish tissue may outweigh the positive health effects of fish consumption in the long term. The estimated daily intake (EDI) of heavy metals ( $\mu$ g/day/person) were calculated (Table 2) in order to estimate the risk of heavy metal toxicity among people who consume fish caught off the Durban coast on a regular basis.

Table 2 shows that the EDI values for As, Ba, Cr, Cu, Pb, and Zn were within the permissible tolerable daily intake (PTDI) for an adult weighing 60.7 kg who consumes an average of 11.66 g of fish a day. Such calculations are necessary to preserve public health because the toxicity of metals is dose-dependent [24]. The value of 11.66 g used for EDI calculations is rather conservative as people are likely to consume more than 11.66 g of fish per day. The advantage, however, with using a conservative figure is that the risk is not overestimated, which may likely cause unnecessary panic among consumers and unwarranted loss of revenue for the fishing industry. However, it still remains a fact that aquatic organisms may bioabsorb toxic metals to levels that might affect even their own physiological state [20].

The EDI for As was 5.09  $\mu$ g/day for an adult African. The main form of As found in fish is the organic As



Figure 2: Log10 average trace metal concentrations in different fish body parts.

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			Metal concentration (µg/g)	ation (µg/g)					
Fish	Fish body part	Cu	Zn	Pb	Cd	Fe	Mn	Cr	Reference
	Muscle	69.3±48.32*	375.66±57.98*	719.01±132.54*	ND	ŊŊ	12.35±0.4*	31.75±16.52*	This study
Cape horse mackerel (Durban coast, South Africa)	Frame	32.9±27*	274.8±28.5*	67.9±59.5*	ND	ND	13.7±1.4*	$18.1\pm1.0*$	
	Gills	54.46±45.35*	283±16.97*	403.1±319.71*	ND	ŊŊ	24.33±5.56*	33.316±16.57*	
(11.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1.1	Muscle	$0.77 \pm 0.14$	$4.21 \pm 0.19$	$0.40 \pm 0.17$	$0.2 \pm 0.02$	$6.25 \pm 0.46$	$0.18\pm0.02$	ND	[20]
norse mackerel (Egypt)	Gills	$2.26 \pm 0.04$	$39.8 \pm 8.16$	$4.03 \pm 1.06$	$0.56\pm0.15$	168.9±38.69	$6.31\pm0.50$	QN	
Canned mackerel (Turkey)	Muscle	$1.01 \pm 0.54$	14.54± 7.31	0.31 ±0.87	$0.02 \pm 0.03$	5.88± 9.44	QN	ND	[27]
Horse mackerel	Muscle	1.6	42	DN	0.04	ND	ND	ND	[28]
Atlantic horse mackerel (Turkey)	Muscle	$0.95 \pm 0.04$	$37.4 \pm 2.9$	$0.68\pm0.05$	$0.50\pm0.03$	74.3 ± 6.1	$7.40 \pm 0.60$	$0.95 \pm 0.07$	[28]
Horse mackerel (Turkey)	Muscle	0.4	7.76	0.0001	ND	8.52	0.58	ND	[29]
Indian mackerel (Persian Gulf)	Muscle	8.86±0.86	35.97±5.33	$6.46 \pm 0.01$	$0.17 \pm 0.06$	ND	ND	ND	[30]
Horse mackerel (southeastern Black Sea)	Muscle	$1.7 \pm 0.04$	$18.1 \pm 0.95$	$0.02 \pm 0.01$	$0.25 \pm 0.03$	ΠN	$0.56\pm0.03$	$0.68\pm0.05$	[31]
ND - Not done *Metal concentration values for the present study were obtained using ash-dried samples	es for the present sti	udv were obtaine	d using ash-dried	samples					

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	Average metal concentration in whole fish (AMC µg/g wet weight)	Daily fish consumption rate (DFC)**	Average adult weight (AAW)*	Estimated daily metal intake***	Permissible tolerable daily intake (PTDI) <sup>a</sup>	PTDI <sup>b</sup> 60.7
Al	117.32	11.66	60.7	13.31	1,000	60,700
As	26.52	11.66	60.7	5.09	1	60.7
Ba	99.33	11.66	60.7	19.08	20	1214
Cr	31.75	11.66	60.7	6.09	5	303.5
Cu	69.3	11.66	60.7	13.32	140	8,498
Mn	12.35	11.66	60.7	2.37	140	8,498
Рb	719.02	11.66	60.7	138.11	3.6	218.52
Zn	375.67	11.66	60.7	72.16	500	30,350
*Averag ***Estin	*Average African adult weight as cited in [15], **DFC value as cite ***Estimated daily metal intake (EDI) calculated as EDI = <u>AMC×DCF</u> *DTD1 values were cited from [16]	**DFC value as cited from [14] as EDI = <u>AMC×DCF</u> AAW				

compounds such as arsenobetaine [25]. While the calculated EDI figure for As may have been lower than the permissible tolerable daily intake for a 60.7 kg adult African (PTDI<sub>607</sub>), which is 60.7  $\mu$ g/day, consistent intake of low levels of As are known to cause adverse health effects, especially skin ailments [26]. If consumed in high quantities, As is known to be acutely toxic and has been linked to an array of ailments including gastrointestinal, cardiovascular and neurological disturbances [25]. Additionally, As has been reported to result in reduced growth and reproduction in both fish and invertebrate populations, and furthermore reduced migration in fish [27]. The type and severity of adverse effects are dependent on the life stages of the fish, with the juveniles more affected than adult fish. Arsenic accumulation may be higher in benthic fish than in epipelagic or photic feeders. However, humans are more sensitive to As than are aquatic organisms, which increases the risk of adverse health effects in the event that contaminated food products are consumed.

The EDI value for Cr was 6.05  $\mu$ g/day, and while this result may suggest insignificant effects on human health, Cr has been observed to variably exert toxic effects at different concentrations in different groups of aquatic organisms, with fish being more resistant compared to invertebrates, especially *daphniids* [28]. For Cu, the EDI was 13.32  $\mu$ g/day compared to the estimated PTDI<sub>60.7</sub> of 8,498  $\mu$ g/day. While Cu is an essential micronutrient that is involved in redox reactions, it is rapidly accumulated by plants and animals and is toxic at low concentrations in water, where early life stages of organisms appear to be more sensitive than adults to copper pollution [27].

In this study, the EDI of Pb as a result of fish consumption was lower than the estimated PDTI value for an adult human. However, regular consumption of fish containing trace levels of bioaccumulated Pb could lead to headaches, abdominal pain, and various symptoms related to the nervous system [25]. For Zn, the EDI as a result of fish consumption was lower than the PDTI<sub>607</sub> value for a grown person (Table 2). Nevertheless, the bioaccumulation of Zn in fish muscle still poses some measure of health risk, especially to regular consumers of fish. A major difference between this study and the findings of Gorur et al. [29] and Gu et al. [30] is that metal concentrations were highest in the flesh (muscle) of the fish and lowest in the gills; however, in this study as opposed to their findings metal concentrations were highest in the gills and liver, and lowest in the muscles of all fish species tested. Our results were against expectation since usually fish are prone to metals adsorbing onto their gill surfaces while a large amount of metallothionein induction is also known to occur in fish liver [29].

#### Conclusion

In conclusion, trace metal pollution of both marine and freshwater resources represents an important environmental and public health concern due to their toxicity and accumulation (bioconcentration) in the food chain. Additionally, these findings point to a need for further and detailed research to determine the extent of metal bioconcentration by different dietary fish species and the extent to which this bioconcentration could be a danger to public health. We suggest that more specific recommendations regarding human consumption are done according to periodic evaluations of levels of environmental pollutants *vis a vis* bioaccumulation levels in dietary fish species.

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