

Original Research

Soil Contamination with Copper and its Effect on Selected Soil Properties After Applying Neutralizing Substances

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Abstract

The aim of this study was to determine the effects of copper contamination (0, 50, 100, 150, and 200 mg Cu/kg of soil) and neutralising substances on the content of total organic carbon, total nitrogen, available forms of phosphorus, potassium, and magnesium, and C:N ratio in soil. The experiment was conducted in four series: without neutralising substances and with compost 3% of soil weight, bentonite 2% of soil weight, and zeolite 2% of soil weight. Contamination of soil with copper and the application of the neutralising substances had a significant effect on soil properties. Contamination with copper in the series without substances increased the content of total nitrogen and available phosphorus, potassium, and especially magnesium. Copper also widened the C:N ratio in soil. All of the amendments applied to the soil had the strongest effect on the content of total nitrogen and the C:N ratio in soil. They reduced the content of total nitrogen and widened the C:N ratio. Amendments also increased the content of available magnesium in soil. The effect of the substances applied to soil on the content of the other elements was significantly weaker, as it changed by up to 12%.

Keywords: soil pollution, compost, bentonite, zeolite, soil properties

Introduction

Growing industrialisation frequently leads to excessive environmental pollution and to irreversible or hardly reversible changes of its status [1-2]. A particular hazard to the soil environment is created by contamination, which is followed by the transfer of contaminants to plants growing on the soils and further parts of the food chain [3]. Trace metals are one of the most hazardous contaminants in the environment.

They change soil properties [4], especially its microbiological and biochemical properties [5], thereby creating disruption in transformations and circulation of elements in the environment [3, 6]. This can lead to limiting the absorption of essential micronutrients and to excessive accumulation of contaminants in crops [7]. This is extremely important in terms of the production of good quality food. Copper is one of the elements that can accumulate in soil in excessive amounts. Its presence in the soil can be a consequence of industrial pollution deposited in the ground, but it can also result from prolonged intensive fertilisation [1]. The content of cobalt in fen soils and limestone soils was higher than in sandy podzolic soils [8]. Copper plays a dual role in

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organisms. It is essential to the life and development of many organisms as it takes part in many physiological processes, including photosynthesis and respiration, usually as a component of many enzymes [9]. Excessive amounts of copper can restrict the growth of plants by having an adverse effect on yield and quality [10], and it causes diseases in humans and animals [11]. Therefore, it is important to maintain the optimum concentrations of copper in soil and to limit its absorption by plants if its level in soil increases excessively [3, 12]. Apart from phytoremediation [13], one of the “*in situ*” methods involves applying different substances in the soil, whose aim is to limit or neutralise the effect of contaminants [14-16]. Organic substances, bentonite, and zeolite are limiting the uptake of trace elements by plants from soil [17]. The sources, forms, mobility and effects of copper contamination and amendments on soil properties were presented in earlier publications [3, 18].

The aim of this study was to determine the effect of copper contamination on the content of total organic carbon, total nitrogen, available forms of phosphorus, potassium, and magnesium, and C:N ratio in soil. The effect of copper on these soil properties was reduced by using compost, bentonite, and zeolite.

Materials and Methods

The study was based on an experiment conducted in the greenhouse at the University of Warmia and Mazury, Olsztyn (northeastern Poland) in 2012. The experiment was carried out on acidic soil with a granulometric composition of sand (sand fraction, 86%; silt fraction, 12%; clay fraction, 2%) according to USDA classification. It had the following properties: pH 5.53 in 1 M KCl; contents of TOC 9.86 g/kg, total nitrogen 0.71 g/kg, C:N ratio 13.89; and contents of available forms of phosphorus 86.48 mg/kg, potassium 108.46 mg/kg, and magnesium 85.32 mg/kg; and content of total copper 4.24 mg/kg of soil. The soil was contaminated with copper (as CaCl_2) at 0, 50, 100, 150, and 200 mg Cu/kg of soil 10 days before sowing. The experiment was carried out in four series: without additives and with compost at 3% of soil weight, bentonite at 2% of soil weight, and zeolite at 2% of soil weight. The concentration of elements in substances was as follows: compost – Corg 204.2 g/kg, total nitrogen 11.97 g/kg, C:N ratio 17.06, P 2.39 g/kg, K 1.41 g/kg, Mg 1.52 g/kg, Cu 0.81 mg/kg; bentonite – P 0.52 g/kg, K 2.36 g/kg, Mg 5.04 g/kg, Cu 21.3 mg/kg; and zeolite – P 0.12 g/kg, K 23.13 g/kg, Mg 0.37 g/kg, Cu 12.38 mg/kg. The choice of the copper doses to the experiment was dictated according to a regulation of the Minister of the Environment of 9 September 2002 on soil and earth quality standards. The amendments were selected based on previous experiments with other trace elements. Moreover, to satisfy the plants' nutritional needs, fertilisers were added to each pot

in the following amounts: 100 mg N [$\text{CO}(\text{NH}_2)_2$], 35 mg P (KH_2PO_4), 100 mg K ($\text{KH}_2\text{PO}_4 + \text{KCl}$), 50 mg Mg ($\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$), 0.33 mg B (H_3BO_3), 5 mg Mn ($\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$), and 5 mg Mo [$(\text{NH}_4)_6\text{Mo}_7\text{O}_{24} \cdot 4\text{H}_2\text{O}$] per kg of soil. Copper chloride, compost, bentonite, zeolite, and fertilisers were thoroughly mixed with 9 kg of soil and transferred into polyethylene pots. In each pot, 8 plants of maize (*Zea mays* L.) San cultivar were sown on 26 April. The moisture content was maintained at 60% of the capillary water capacity during the experiment. After 82 days since the start of the experiment (16 July), soil samples for laboratory tests were taken during the maize harvest in the end of tassel emergence – BBCH scale. The experiment was realised in three repetitions (60 pots).

The soil samples were dried and sifted through a 1 mm mesh sieve. Before starting the experiment the soil, compost, bentonite, and zeolite were mineralized in concentrated nitric acid (HNO_3 , analytically pure) in a MARS 5 microwave oven (CEM Corporation, USA), by the US-EPA3051 method [19]. Content of total copper was determined by flame atomic absorption spectrometry (FAAS). The soil pH in 1 mol/dm³ KCl was determined potentiometrically [20]. The following were determined in the samples (in three replications): organic carbon by the Tiurin method [20], total nitrogen by the Kjeldahl method [20], available forms of phosphorus and potassium by the Egner-Riehm method, and magnesium (Mg) by the Schachtschabel method [20]. Wet mineralisation of the soil samples for total nitrogen was carried out in concentrated sulphuric acid (H_2SO_4 , analytically pure). Next, it was analysed in a Buchi K-355 Kjeldahl Distiller. The results were analysed statistically with the STATISTICA software package and a two-way analysis of variance ANOVA. Moreover, the effect of contamination with copper and the impact of compost, bentonite, and zeolite on soil properties was assessed by principal component analysis (PCA) and with Pearson's correlation coefficients [21].

Results and Discussion

Contamination of the soil with copper and the application of neutralising substances had a significant effect on soil properties (Tables 1-2). Contamination of the soil in the series without amendments increased the content of total nitrogen ($r = 0.441$) and available phosphorus ($r = 0.808$), potassium ($r = 0.918$), and, especially, magnesium ($r = 0.891$). It also narrowed the C:N ratio ($r = -0.441$) in soil. The increase in the content of available magnesium in the soil contaminated with 200 mg Cu/kg of soil was 49% and it was greater than that of available potassium (33%), total nitrogen (24%), and available phosphorus (20%), compared to the control (no copper). The organic carbon-to-total nitrogen ratio in the same pot was reduced by 24%.

The greatest effect of all substances was observed in the content of total nitrogen and the C:N ratio in soil

Table 1. Effect of copper contamination and neutralizing substances on content of organic carbon, total nitrogen, and C:N ratio in soil.

| Copper dose (mg/kg of soil) | Without amendments | Compost | Bentonite | Zeolite | Average |
|-----------------------------|---|-----------|-----------|-----------|----------|
| Organic carbon (g/kg d.m.) | | | | | |
| 0 | 9.83±0.25 | 9.15±0.05 | 8.93±0.11 | 8.03±0.11 | 8.99 |
| 50 | 9.75±0.07 | 9.90±0.22 | 8.70±0.22 | 8.55±0.21 | 9.23 |
| 100 | 9.45±0.21 | 9.08±0.11 | 9.00±0.05 | 8.55±0.08 | 9.02 |
| 150 | 9.00±0.22 | 9.45±0.21 | 8.10±0.12 | 8.85±0.21 | 8.85 |
| 200 | 9.30±0.05 | 8.85±0.21 | 7.80±0.07 | 8.03±0.23 | 8.50 |
| Average | 9.47 | 9.29 | 8.51 | 8.40 | |
| r | -0.846** | -0.410 | -0.853** | 0.131 | -0.792** |
| LSD for: | Cu dose - n.s., substances application - 0.53**, interaction - n.s** | | | | |
| Total nitrogen (g/kg d.m.) | | | | | |
| 0 | 0.80±0.05 | 0.48±0.06 | 0.60±0.04 | 0.50±0.01 | 0.60 |
| 50 | 1.10±0.07 | 0.63±0.03 | 0.44±0.01 | 0.52±0.02 | 0.67 |
| 100 | 1.13±0.02 | 0.68±0.06 | 0.51±0.03 | 0.53±0.03 | 0.71 |
| 150 | 1.10±0.02 | 0.53±0.02 | 0.66±0.02 | 0.55±0.05 | 0.71 |
| 200 | 0.99±0.05 | 0.52±0.01 | 0.47±0.04 | 0.56±0.06 | 0.64 |
| Average | 1.02 | 0.57 | 0.54 | 0.53 | |
| r | 0.441 | -0.038 | -0.069 | 0.993** | 0.369 |
| LSD for: | Cu dose - 0.04**, substances application - 0.05**, interaction - 0.09** | | | | |
| C:N ratio | | | | | |
| 0 | 12.29 | 19.06 | 14.88 | 16.06 | 15.57 |
| 50 | 8.86 | 15.71 | 19.77 | 16.44 | 15.20 |
| 100 | 8.36 | 13.35 | 17.65 | 16.13 | 13.87 |
| 150 | 8.18 | 17.83 | 12.27 | 16.09 | 13.59 |
| 200 | 9.39 | 17.02 | 16.60 | 14.34 | 14.34 |
| Average | 9.42 | 16.60 | 16.23 | 15.81 | |
| r | -0.612* | -0.143 | -0.227 | -0.876** | -0.760** |
| LSD for: | Cu dose - 1.25**, substances application - 1.40**, interaction - 2.80** | | | | |

Values are average ± standard deviation; significant at ** $P \leq 0.01$ * $P \leq 0.05$; n.s. - non-significant; r - correlation coefficient

(Tables 1-2). Compost, bentonite, and zeolite decreased the average content of total nitrogen by 45%, 48%, and 48% and widened the C:N ratio by 76%, 72%, and 68%, respectively, compared to the series without additions. They also increased the content of available magnesium, which ranged from 11% (compost) to 22% (bentonite) compared to the soils without amendments. The effect of the substances on the content of total organic carbon and available phosphorus and potassium was much smaller than on the content of total nitrogen, magnesium, and C:N ratio. The trend of changes of these elements depended on the kind of substance. Zeolite reduced the content of organic carbon, available phosphorus, and potassium by 11-12% and bentonite,

and that of organic carbon by 10% compared to the series with no additions. Compost increased the average content of available phosphorus by 6% and that of potassium by 12%. Bentonite increased the content of available potassium by 8%.

The PCA (Fig. 1) and the correlation coefficients (Table 3) revealed the significant relationship between the content of some elements and the C:N ratio, acidity, and other soil properties. Significant positive correlations were found to exist between:

- The content of total nitrogen and organic carbon and hydrolytic acidity of soil.
- The content of available phosphorus and potassium and the soil hydrolytic acidity.

Table 2. Effect of copper contamination and neutralizing substances on content of available phosphorus, potassium, and magnesium in soil.

| Copper dose (mg/kg of soil) | Without amendments | Compost | Bentonite | Zeolite | Average |
|-----------------------------------|--|-----------|-----------|-----------|---------|
| Available phosphorus (mg/kg d.m.) | | | | | |
| 0 | 98.8±2.0 | 99.0±2.1 | 90.5±1.5 | 91.9±2.3 | 95.1 |
| 50 | 84.8±2.2 | 116.9±0.7 | 95.9±1.9 | 87.8±1.4 | 96.4 |
| 100 | 99.5±0.4 | 113.9±0.2 | 95.9±1.2 | 98.6±2.5 | 102.0 |
| 150 | 112.6±0.3 | 111.0±0.8 | 101.3±1.9 | 95.3±2.0 | 105.1 |
| 200 | 118.7±2.0 | 106.2±1.5 | 112.8±1.6 | 83.4±1.7 | 105.3 |
| Average | 102.9 | 109.4 | 99.3 | 91.4 | |
| r | 0.808** | 0.191 | 0.934** | -0.951** | 0.959** |
| LSD for: | Cu dose - 7.39**, substances application - 8.26**, interaction - 16.52* | | | | |
| Available potassium (mg/kg d.m.) | | | | | |
| 0 | 128.7±2.1 | 137.2±3.0 | 162.6±3.0 | 99.1±3.0 | 131.9 |
| 50 | 124.5±3.0 | 141.4±4.0 | 132.9±2.0 | 107.6±3.0 | 126.6 |
| 100 | 137.2±2.6 | 145.6±0.9 | 179.5±0.8 | 128.7±1.2 | 147.8 |
| 150 | 147.7±3.0 | 175.2±3.0 | 137.2±1.2 | 132.9±3.0 | 148.3 |
| 200 | 171.0±3.5 | 192.2±3.0 | 149.9±3.0 | 158.3±2.0 | 167.9 |
| Average | 141.8 | 158.3 | 152.4 | 125.3 | |
| r | 0.918** | 0.943** | -0.175 | 0.924** | 0.913** |
| LSD for: | Cu dose - 7.69**, substances application - 8.60**, interaction - 17.20** | | | | |
| Available magnesium (mg/kg d.m.) | | | | | |
| 0 | 92.3±0.6 | 129.1±1.1 | 131.2±2.2 | 95.7±2.9 | 112.1 |
| 50 | 88.5±0.8 | 113.7±2.7 | 131.2±2.1 | 129.1±3.0 | 115.6 |
| 100 | 92.7±2.1 | 114.1±2.1 | 148.3±2.1 | 135.5±3.1 | 122.7 |
| 150 | 127.0±3.1 | 122.7±2.1 | 122.7±4.2 | 135.5±4.2 | 127.0 |
| 200 | 137.6±3.0 | 119.3±4.2 | 122.7±2.4 | 125.2±3.9 | 126.2 |
| Average | 107.6 | 119.8 | 131.2 | 124.2 | |
| r | 0.891** | -0.261 | -0.386 | -0.866** | 0.951** |
| LSD for: | Cu dose - 12.96**, substances application - 14.22*, interaction - n.s. | | | | |

Values are average ± standard deviation; significant at ** $P \leq 0.01$ * $P \leq 0.05$; n.s. - non-significant; r - correlation coefficient.

- The content of available potassium and the cation exchange capacity.
 - The content of available magnesium and C:N ratio.
- Significant negative correlations were found to exist between:
- Total nitrogen and available magnesium.
 - Total nitrogen and C:N ratio.
 - The content of available phosphorus and pH and base saturation.
 - The content of available magnesium and soil pH.
 - The C:N ratio and soil hydrolytic acidity (Table 3).

The relationships between the soil macroelements, especially between total nitrogen content, available

phosphorus, potassium, and magnesium, and the C:N ratio were confirmed by PCA analysis (Fig. 1). In addition, the vectors perfectly reproduced the contents of available magnesium, total nitrogen, hydrolytic acidity, and base saturation, and it poorly reproduced the C:N ratio – especially total organic carbon. The principal components represented in total 57.88% (including the first component at 30.15% and the second at 27.73%). The data dispersion in Fig. 2 confirms the effects of compost, manure, and zeolite on soil properties, although it is difficult to claim beyond doubt which of these substances had the greatest effect.

Table 3. Correlation coefficients (r) between content of elements and C:N ratio and some properties of soil.

| Factor | C _{org} | Total-N | C:N ratio | P | K | Mg |
|-------------------|------------------|---------|-----------|---------|---------|----------|
| pH _{KCl} | 0.084 | 0.210 | -0.230 | -0.367* | -0.292 | -0.461** |
| HAC | 0.061 | 0.365* | -0.380* | 0.345* | 0.457** | 0.186 |
| TEB | -0.287 | -0.143 | -0.013 | -0.193 | 0.266 | 0.272 |
| CEC | -0.264 | -0.058 | -0.096 | -0.110 | 0.358* | 0.304 |
| BS | -0.221 | -0.376* | 0.290 | -0.380* | -0.219 | 0.023 |
| C _{org} | | 0.411** | -0.137 | 0.250 | 0.097 | -0.127 |
| Total-N | | | -0.930** | 0.127 | 0.035 | -0.337* |
| C:N ratio | | | | -0.075 | -0.050 | 0.331* |
| P | | | | | 0.371* | 0.055 |
| K | | | | | | 0.266 |

pH_{KCl} - pH in 1M KCl, HAC - hydrolytic acidity, TEB - total exchangeable bases, CEC - cation exchange capacity, BS - base saturation; significant at ** $P \leq 0.01$ * $P \leq 0.05$; r - correlation coefficient

Excessive amounts of copper in soil can disrupt the process of nutrient absorption by plants [3]. Since copper is firmly fixed in soil by organic substances, silty substances, and hydrated iron and aluminium oxides, it is important to increase the content of organic substances in soil as it has the greatest impact on its sorptive properties [22]. The application of organic and mineral substances with strong sorptive properties and adjusting soil pH reduces the bioavailability of copper and its toxicity to plants and soil microorganisms and increases the bioavailability of macroelements [14-15]. The correlation between pH and sorptive properties of soil and content of elements in soil was partly confirmed in our study. Particularly good results can be achieved

by applying organic matter to soil. The application of elements with organic substances caused an increase in these elements in soil. Bednarek et al. [23] used liquid manure and found the content of available phosphorus, potassium, and magnesium and organic carbon in soil to increase. According to Wright et al. [24], compost increases the content of phosphorus, potassium, and magnesium in soil. The application of compost to soil in an experiment carried out by Wyszowski and Sivitskaya [25] increased the accumulation of organic carbon and available potassium in soil. Kondratowicz-Maciejewska and Kobierski [26] found that manure increased the content of available phosphorus, potassium, and magnesium, and its content was correlated to the content of organic carbon in soil, which was partly confirmed

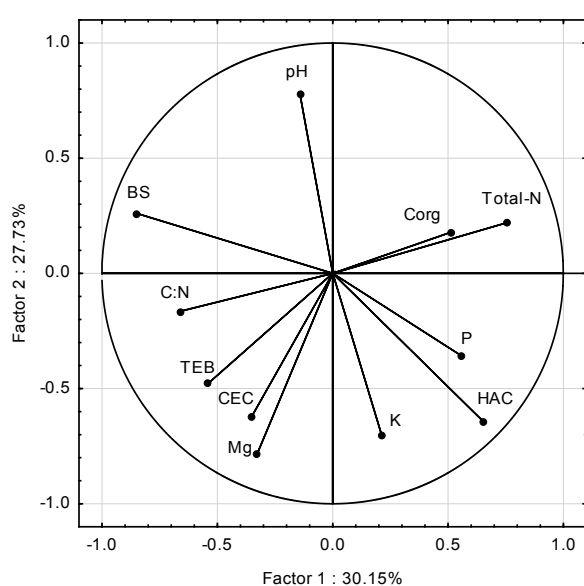


Fig. 1. Content of elements and C:N ratio in the soils illustrated with the PCA method; key: vectors represent analyzed variable (content of C_{org}, total-N, available P, K, Mg, and C:N ratio).

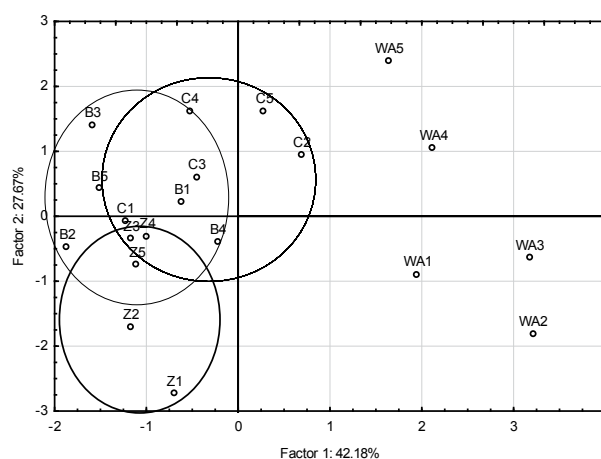


Fig. 2. Effect of neutralizing substances on content of macroelements and C:N ratio in the soils illustrated with the PCA method; key: points show soil samples with elements and C:N ratio (WA - without amendments, C - compost, B - bentonite, Z - zeolite; 1 - 0, 2 - 50 mg, 3 - 100 mg, 4 - 150 mg, 5 - 200 mg/kg of soil).

in this study. An increase of elements in soil contaminated with the highest doses of copper was correlated with the lowest maize yield. The uptake of elements by maize from soil contaminated with copper was lower than from control soil (without the application of copper).

Furthermore, Šimanský and Kováčik [27] found the content of organic carbon to be positively correlated to hydrolytic acidity and negatively correlated to the base saturation, which was not confirmed in this study. Moreover, as in this study, Kondratowicz-Maciejewska and Kobierski [26] found the content of available potassium to be correlated positively to the content of available magnesium. Wyszkowski and Sivitskaya [25] found bentonite to have a greater effect on the content of available forms of nutrients in soil than compost and zeolite. Bentonite increased the content of available phosphorus, potassium, and magnesium and zeolite only increased the content of available magnesium. Wyszkowski and Modrzewska [28] found the use of bentonite and zeolite to have a positive effect on the content of organic carbon, available phosphorus, and the C:N ratio in soil. Moreover, bentonite increased the content of available potassium in soil. According to Ostrowska et al. [29], intensive humification and mineralisation of organic matter and an increase of elements in soil with C:N ratio lower than 10, which was confirmed in this study in objects with copper contamination. A wide C:N ratio has a negative effect on nitrogen content in soil because of uptake of this and another elements from soil solution. These relationships were observed in our study.

Conclusions

Contamination of soil with copper and the application of neutralizing substances had a significant effect on soil properties. Contamination with copper in the series without substances increased the content of total nitrogen and available phosphorus, potassium, and especially magnesium. Copper also widened the C:N ratio in soil. All of the amendments applied to the soil had the strongest effect on the content of total nitrogen and the C:N ratio in soil. They reduced the content of total nitrogen and widened the C:N ratio. Amendments also increased the content of available magnesium in soil. The effect of the substances applied to soil on the content of the other elements was significantly weaker, as it changed by up to 12%.

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Conflict of Interest

The authors declare no conflict of interest.

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