Original Research

Role of River-Flow in Alleviating Heavy Metal Load from a Polluted Urban River

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Abstract

Optimum water flow is inevitable for river health protection. Urban rivers around Dhaka city, Bangladesh, is heavily polluted by numerous chemical pollutants including heavy metals. This study investigated heavy metals concentration and river-flow volume to elucidate the relationship between them. River-flow and metals analysis (n = 672 water samples) was performed for two consecutive years (November, 2017- October, 2019) in both the dry- and wet-seasons. The metal was analyzed by AAS, and Water-flow was measured by the formula of 'velocity dimensional'. It was observed that, in the dry-period (November to May), the water-flow of the river ranged from 12.76 m³/sec to 49.87 m³/sec, whereas, in the wet-period (June to October), it was 35.39 m³/sec to 183.48 m³/sec. In the dry-period, the concentration of Cr, Cd, Pb, and Ni was found 3,860%, 12,900%, 8,200%, and 3,300%, respectively higher than the permissible limits. Optimum water-flow, \geq 49.87 m³/sec, can run-down these metals from the river both in the dry- and wet-periods to retain environmental balance throughout the year. The proposed modeling of the optimum water-flow to minimize heavy metals concentration would be a promising measure to tackle the risk of the world's urban rivers.

Keywords: communal method, metal concentration, optimum flow, urban river, water-flow

Introduction

Environmental flow is required to ensure the ecosystem, as well as the survival of aquatic life in the urban river [1]. However, rivers in urban areas around the world continued to be polluted by untreated

wastewater from industrial and domestic sources. As a result, the degree of physicochemical parameters, especially the concentration of heavy metals (HMs) in urban rivers, has reached alarming levels [2, 3]. Buriganga and Turag rivers of Bangladesh, Citarum River of Indonesia, Sarno River of Italy, Pasing River of Manila, Cuyahuga River of USA, and Buddha Nulla River of India, etc. are the best possible examples of urban polluted rivers.

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Heavy metals (HMs) are dangerous because they tend to bio-accumulate. Since the toxicity of HMs precipitates over time, an immediate regression of their concentration is essential [4, 5]. Aside from the environmental flow, a "certain amount" of the water flow can inhibit metals from forming poorly soluble inorganic compounds and burying them in sediments. Therefore, the water flow can keep a vital role in removing metal concentration. However, with the loss of water flow, the worst level of contamination is generally found in urban rivers [6].

Contamination of HMs in river water has been found at an alarming level [7, 8] because untreated wastewater sources are increasing with unplanned urbanization, population growth, and massive industrialization around the urban rivers. The concentration of HMs in river water can pose adverse risks to the ecosystem; [9, 10, 11]. The excessive use of HMs in industries around the river bank poses a greater risk of pollution of river water. Thus, HM concentrations in the water can continue to increase.

Many urban rivers around the world have been exposed to HMs contamination in their sediment and basins, surface water, and in the food chain. For example, the water quality and seasonal variations of Nile River of Egypt and Odiel River of South-West Spain was assessed in terms of HMs concentration [12, 13]. The existence of Zn, Fe, Mn, Cu, Cd, Pb, and As was confirmed with much lower quantities. Similar variations have also been studied for the Siahroud River of Iran [14], Indus River of Pakistan [15] and the Bramani River of India [16]. With a different analysis direction, a study was conducted to find out the concentration, stability, risk assessment, and source apportionment of HMs of the Songhua River in the Harbin region of Northeast China [17]. Another study demonstrated the contamination of Buriganga in Bangladesh [18].

However, to improve the water quality of urban rivers, many developed countries have implemented aeration, water diversion, sediment dredging, and flocculation methods [19]. For example, the river aeration technology has been used in the Oerias River in Portugal, the Emsche River in Germany, the Thames River in the UK and the Homewood canal in the United States [20]. Moreover, sediment dredging has been implemented into many rivers around the world, such as the UK port-channel and Xuan Wu Lake in China [21]. In addition to the technology used in developed countries, retaining an optimum water-flow in urban rivers can be an option to minimize the concentration of heavy metals in urban rivers.

This paper, therefore, elucidated the water-flow and a set of HMs in Turag river, which is one of the most polluted rivers in and around Dhaka city, to find out optimum water-flow, which will run-down HMs of the river to retain sustainable water quality.

Materials and Methods

Study Area

Bangladesh is a very small but one of the most populated riverine countries in the world. The Turag is one of the most polluted urban rivers, surrounded by numerous industries and population growth. Department of Environment (DoE), Bangladesh, has declared the river as ECA (Ecologically Critical Area) to save its ecology in September 2009 [22]. A 4 km long strip of river Turag was investigated where sampling sites (SS) were selected based on the bend and width of the river, population growth, surrounding industries, farmland, and slum (Fig. 1). The distance from one SS to another SS was 1 km. The width of the river (in the study area) was in the range of 20 to 25 m. The depths vary from 5 m to 3 m in the wet to dry season. Zone temperature varies from 16°C to 36°C. Longitude, latitude, and distance from pollution sources of the area are also illustrated in Table 1.



Fig. 1. Area of study and the locations of sample collection in Turag River, Bangladesh (courtesy: Google map).

Location	Longitude	Latitude	Distance*
SS-1	90º24′3.389′′ E	23º52′54.65′′ N	20 m
SS-2	90º23´32.51´´ E	23º53′11.09′′ N	1000 m
SS-3	90º23´23.43´´ E	23º53′29.61′′ N	524 m
SS-4	90º23′10.51′′ E	23º53′50.08′′ N	43 m

Table 1. Study area specifications.

*Minimum distance from the nearby industrial areas (measured from the Google map)

Pollution Sources

In general, the pollution point sources of Turag River are mainly industrial effluents and municipal waste, and the non-point sources are agricultural runoff and others. Industrial effluents are mostly textile, tannery, dyeing, and pharmaceuticals. DoE & LGED. 2010 reported that the Turag River received discharges from all the industries that include more than 23 heavy industries including textiles, footwear, food, chemical, pharmaceutical, and detergents. This cluster of industries generates daily more than 7,000 kg of effluents [23].

Sample Collection, Preservation, and Analysis

A total of 28 water samples were collected each month (from November 2017 to October 2019) from four sampling sites, i.e., two samples from each bank and three samples from the middle of the river for each site (see small dots in Fig. 1). A seasonal study was conducted in the dry and wet periods (November-May: the dry period, June-October: the wet period) of each year. Therefore, a total of 672 samples (i.e., 336 samples per year) were collected. Water-flow was measured by the formula 'dimensions of velocity' (Eq. 1) using technical support from the Bangladesh Water Development Board.

$$V = 0.25639 N/100 + 0.00339 m^{3}/sec$$
 (1)

...where, V = velocity and N = number of revolution

The water samples were collected in 500 ml plastic bottles, which were sterilized before sample collection to avoid contamination. They were cleaned with a detergent solution followed by overnight treatment with 5%, and a final wash using de-ionized water repeated a couple of times. After the collection of samples, those were sealed and preserved at to prevent the risk of hydrolysis and oxidation. The bottles were kept airtight and labeled individually for documentation. The sample-water in each bottle was digested with nitric acid and filtered by Whatman-41 (0.45 μ m pore size) filter paper for a guesstimate of dissolved metal content. Moreover, it was kept at room temperature with 2 ml nitric acid to resist the precipitation of metals. A test volume of 50 ml sample was transferred into a clean glass, and concentrated was added for its digestion. The temperature was kept to heat the solution until dry without boiling. After cooling down, distilled water was used to dilute the samples and made a volume of 50 ml in a volumetric flask. Finally, the samples were aspirated into an air acetylene flame in the AAS (Shimadzu AA6300) of the Bangladesh Atomic Energy Research Establishment (AERE).

The metal content, MC of the sample, is finally determined using Eq. (2), where C is the concentration of the metal determined by the AAS, and W is the weight of the sample. The numerical data of the HM concentration and water-flow were statistically processed, analyzed, and represented using the MATLAB application (R2016b v.9.1.0441655).

$$MC = C/W \times 100 \text{ (ppm)} \tag{2}$$

Results and Discussion

Impact of Cumulative Heavy Metals Concentration

The present study has recorded progressively higher concentrations of the HMs compared to the earlier studies. Thus, an imminent endangerment to aquatic life and the territorial environment has been revealed. More quantitative details of the concentration levels are given in Table 2 and Table 3. The assessed amount of Cr, Cd, Pb, and Ni is found to be 3,860%, 12,900%, 8,200%, and 3,300% higher than the limits of Bangladesh Standard [24].

Compared with Bangladesh standard of the respective heavy metals, it was observed that concentrations of heavy metal had reached a critical state. For example, in 2017-2018, the highest value of Fe was found at 3.826 ppm in March 2018. Contrariwise, in 2018-2019, the highest concentration of Fe was found 3.390 ppm in April 2019 (Fig 2). It is evident that the highest concentration was found in the dry period of both years. The concentrations of the Cu and Zn were within the permissible limit in the entire period of our study.



Fig. 2. Impact of water-flow on HM concentrations of Turag River in the year of a) 2017-18 and b) 2018-2019.

Impact of Plummeting Water-Flow

The monthly water flow for the study period has been shown in Table 2 and in Table 3. Compared to the previous cases, it has been found that the river flow has decreased sharply in the last two decades. For example, see Table 4, the lowest and highest water flows were recorded at 180.77 m3/sec and 556.85 m³/sec, respectively, in 1989-2004. In the next five years, the lowest flow decreases by 65.8% and the highest flow decrease by about 16%. In 2016-2017, the volume of water flow decreased to 6.99 m3/sec in the dry period. This means that the rate of water flow reduction is greater than 80% of the initial water flow measured in 1989-2004. However, the lowest and highest water flow in the current study showed a very miserable condition of 12.76 m³/sec and 158.47 m³/sec respectively.

The HMs concentration and the water-flow are reversely found (Fig. 2). In a dry season, the flow-rate of the river naturally reduces. However, it increases in wet seasons. In contrast, the concentration of HMs has a declining trend in the wet season and vice-versa. For example, the highest Cd-concentration was found in May 2018 (0.629 ppm), when the water flow was 33.82 m³/sec, although it was gradually reduced with increasing water-flow up to 158.47 m³/sec. The Cd-

concentration was found Below the Detection Level (BDL) in September 2018, when the water-flow was 158.47 m^3 /sec. A similar effect of water-flow exists on other HMs.

With another example, in Fig. 2, it was observed that the highest concentration of Pb was found 0.818 ppm in February 2019 in the dry period, when the water-flow was 24.82 m³/sec. The value was higher than the Bangladesh standard (0.5 ppm) like Cd. In the wet period, the concentration was BDL. This trend in the concentration holds for other metals as well.

Analysis of Concentration of HMs with Water-Flow

In Fig. 3, the average concentration of the metals was plotted over the average water-flow, where the flow is normalized by a mean of 44.02 and a standard deviation of 55.02. An analytical relationship between concentration and water-flow is estimated in Eq. (2) at 95% confidence interval using MATLAB polynomial curve-fitting toolbox. We found that a fifth-degree polynomial would reasonably capture the empirical relation between the HMs concentration and water-flow. We computed the flow-coefficients (P) in Table 5 for this model utilizing the HMs concentration and water-flow evaluated in our study. Thereby,

Table 2. Monthly	water-flow :	and heavy metals concentra	tion of Turag Rive	r (dry period).	-			-		
Month	Period	Water Flow (m ³ /sec)	Cr (ppm)	Cd (ppm)	Pb (ppm)	Cu (ppm)	Mn (ppm)	Fe (ppm)	Zn (ppm)	Ni (ppm)
Nov 17		13.770	0.091 ± 0.015	0.060 ± 0.010	BDL	0.018 ± 0.005	0.098 ± 0.010	0.320±0.025	0.090±0.005	0.031 ± 0.025
Dec 17		20.102	0.206 ± 0.010	0.075 ± 0.002	0.008 ± 0.010	BDL	$0.367^*\pm 0.035$	1.013 ± 0.010	0.134 ± 0.025	0.042 ± 0.001
Jan 18		17.453	$2.010^*\pm0.020$	0.217 ± 0.030	0.424 ± 0.025	0.038±0.002	$0.950^*\pm 0.002$	$1.638^{*}\pm0.035$	0.411 ± 0.030	$0.654^*\pm 0.001$
Feb 18	Dry	21.733	2.390*±0.025	$0.648^*\pm0.005$	0.054 ± 0.001	0.092 ± 0.010	$1.448^{*}\pm 0.025$	3.674*±0.040	0.605±0.005	$0.694^*\pm 0.005$
Mar 18		12.769	$3.568^{\pm}0.020$	$0.593^{*}\pm0.035$	$0.513^{*}\pm0.002$	0.054±0.005	$1.834^*\pm 0.010$	$3.826^*\pm 0.001$	0.680 ± 0.040	$0.926^*\pm 0.010$
Apr 18		25.468	3.565*±0.021	$0.546^*\pm 0.005$	0.056 ± 0.025	0.053±0.005	2.057*±0.025	$1.501^*\pm 0.040$	0.775±0.015	$0.229^*\pm0.035$
May 18		33.820	$0.339^*\pm 0.030$	0.429 ± 0.005	0.094 ± 0.040	0.006 ± 0.010	0.246 ± 0.040	$3.149^{*}\pm0.010$	0.629 ± 0.010	0.064 ± 0.005
Nov 18		49.874	0.095 ± 0.001	0.005 ± 0.030	BDL	0.126±0.025	0.099±0.025	0.911±0.025	0.091 ± 0.005	BDL
Dec 18		33.132	0.218±0.015	0.327 ± 0.030	BDL	0.060 ± 0.010	0.267±0.030	1.058 ± 0.010	0.149±0.002	0.063 ± 0.005
Jan 19		27.642	$2.268^{*}\pm0.020$	0.390 ± 0.005	0.023 ± 0.025	0.058±0.025	$1.448^*\pm 0.010$	2.722*±0.040	0.613±0.010	$0.498^*\pm 0.030$
Feb 19	Dry	24.821	2.250*±0.025	$0.609^*\pm0.040$	0.018 ± 0.001	0.034±0.005	$1.239^{*}\pm0.002$	2.793*±0.025	0.842 ± 0.030	$0.944^*\pm 0.001$
Mar 19		23.156	2.005*±0.015	$0.609^*\pm 0.001$	0.057 ± 0.025	0.009±0.005	$1.883^{*}\pm0.030$	$2.607^{*}\pm0.001$	0.822±0.010	$1.327^{*}\pm0.025$
Apr 19		25.623	$3.617^{*}\pm0.001$	$0.556^*\pm0.035$	0.089 ± 0.010	0.046 ± 0.010	$1.891^*\pm 0.010$	$3.390^*\pm 0.025$	0.981 ± 0.002	$0.766^* \pm 0.010$
May 19		23.236	$1.921^*\pm 0.010$	$0.536^*\pm 0.025$	0.080 ± 0.002	0.003±0.001	$1.518^*\pm 0.035$	$2.838^{*}\pm0.010$	0.737±0.025	$0.930^*\pm 0.025$

BDL: Below Detection Level *Exceeded Bangladesh standard limit (EQS 1997)

	$Mn \ (ppm) \qquad Fe \ (ppm) \qquad Zn \ (ppm) \qquad Ni \ (ppm) \qquad \\$	0.058±0.010 0.731±0.005 0.474±0.002 0.317±0.035	0.011±0.002 0.875±0.040 0.363±0.030 0.016±0.005	0.092±0.025 0.596±0.002 BDL 0.027±0.002	0.090±0.010 0.591±0.005 BDL BDL BDL	$0.096\pm0.035 \qquad 0.671\pm0.040 \qquad 0.088\pm0.002 \qquad 0.025\pm0.001$	0.013±0.001 0.637±0.025 0.598±0.025 BDL	0.080±0.010 0.464±0.030 0.211±0.010 BDL	0.096±0.002 0.341±0.010 BDL 0.028±0.010	0.091±0.025 0.275±0.002 0.134±0.005 0.026±0.002	0.092±0.010 0.336±0.025 0.090±0.035 0.019±0.010	
	Cu (ppm)	0.024±0.002	BDL	BDL	BDL	0.016±0.010	0.029±0.025	0.022±0.010	0.014±0.002	0.012±0.025	0.012±0.010	
	Pb (ppm)	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	
ver (wet period).	Cd (ppm)	0.369±0.005	0.168±0.002	0.053±0.001	BDL	0.055±0.002	0.318 ± 0.040	0.241±0.005	BDL	0.100 ± 0.002	0.053 ± 0.025	
ration of Turag Ri	Cr (ppm)	0.856 ± 0.024	0.134 ± 0.015	BDL	BDL	0.034 ± 0.003	0.402±0.015	0.070±0.002	BDL	0.068 ± 0.030	0.120 ± 0.035	
and heavy metals concent	Water Flow (m ³ /sec)	61.880	115.51	150.92	158.47	106.43	97.405	122.11	140.25	120.07	121.16	
y water-flow a	Period			Wet					Wet			Jatantion I are
Table 3. Monthl	Month	Jun 18	Jul 18	Aug 18	Sep 18	Oct 18	Jun 19	Jul 19	Aug 19	Sep 19	Oct 19	BDI - Balow F

the minimum required water-flows for the permissible concentrations of individual metals are determined.

$$C = P_1 * f^4 + P_2 * f^3 + P_3 * f^2 + P_4 * f + P_5 \quad (3)$$

The minimum required water-flow is determined in two steps. First, the individual concentrations of the metals are determined to their known permissible limits. For example, see Fig. 3, the required minimum flow for Cr was found 17.90 m³/sec for the permissible concentration of 0.5 ppm. Thus, the minimum required flows found for other metals are 7.82, 21.5, 50.0, 33.2, and 44.30 m³/sec for Cd, Pb, Mn, Fe, and Ni, respectively. Therefore, according to the graphical presentation, when the river could maintain its waterflow of more than 50 m3/sec, manganese present in the river will reach into an allowable limit. Since all the results found in the illustration were less than 50 m³/sec, it is obvious that a water-flow of 50 m³/sec is also able to reduce the concentration of other HMs to the permissible limit, which supports the experimental findings of optimum required water-flow into the Turag River. Note that the minimum required flows of Cu and Zn were not found, as their concentrations always remained within permissible limits.

Discussion

To the best of our knowledge, this is the first study that examined the role of river-flow on HMs load reduction quantitatively in Bangladesh. The rapid expansion of urbanization and industrialization has an impact on river flow regime along with the other hydro geomorphological factors. In this study, we hypothetically took river-flow as a pivotal factor to reduce HM load in the urban river to examine through field data analysis for two consecutive years. Usually, during the course of HM transportation, it undergoes a set of processes including speciation, deposition, dissolution, precipitation, and sorption [25]. Not surprisingly, when comparing with the other urban rivers around the world, it was observed that most of the considered HMs are present in the Turag's water at a much higher level. Unlike the Odiel River in Spain, where Zn, Cu, Mn, and Fe were recorded at their highest levels, Turag recorded the secondhighest levels of all the metals. High loading of HMs negatively impacts on the river ecosystem health. The Government of Bangladesh has been looking for a suitable way to alleviate the pollution from the urban river for ecosystem restoration. But the outcome of the restoration program is not visible yet due to multilateral factors. Under the circumstance, natural river-flow regime re/installation could be a suitable option to achieve a large-scale pollution abatement goal. Therefore, this experimental study might shed light on the crucial role of river flow to remediate HM pollution load through dissolution phenomena. Here, to find out the optimum flow of water, initially, the optimum flow



Fig. 3. Analytical presentation of concentration (C) in ppm of (a) Cr, (b) Cd, (c) Pb, (d) Cu, (e) Mn, (f) Fe, (g) Zn and (h) Ni with water flow in m³/sec.

of an individual metal is determined via statistical modeling (Fig. 3). Thereafter, the overall optimum flow is analytically determined as its highest value. For example, as mentioned in Table 6, the highest of the individual minimum flow is 49.87 m³/sec (for Fe). Thus,

the optimum flow of 49.87 m³/sec is a discretionary requirement in general for all metals to run-down and remain within permissible limits. Practically it was also observed that all the metals remain within their limits with that flow, as evident in Table 2 (Nov 18).

Duration	Water Flow (m ³ /sec)					
Duration	Minimum	Maximum				
1989-2004 [1]	180.77ª	556.85 ^b				
2005-2009 [1]	62.33ª	467.82 ^b				
2016-2017 [6]	6.99ª	93.05 ^b				
2017-2019 (Present Study)	25.75ª	119.42 ^b				

Table 4. Current state of the water-flow in the Turag River.

^aAverage value in the dry periods

^bAverage value in the wet periods

A line of evidence from a global database represents the role of high river-flow on HMs load reduction for instance, the concentration of heavy metals in the Nile was found to be within the permissible limit while the water-flow was 84 bcm/yr (over 49.87 m³/sec). The concentration of heavy metals in the Brahmini River was observed within the WHO standard limit, while the water flow was found to be 19,457 m³/sec in the wet period (over 49.87 m³/sec). Likewise, being one of the most polluted rivers in the world, heavy metal concentration of Citarum River, Indonesia, has been found within acceptable limit due to the maintenance of water-flow of 78.74 m³/sec [26]. On the other hand, HMs concentration of Sarno River, Italy, the most polluted river of Europe [27], was found higher than the standard limit owing to very low water-flow (13 m³/sec) throughout the year. Similarly, a very low flow of water (12 m³/sec) from Pasig River, Manila, makes it one of the most contaminated rivers in the world [28]. Therefore, it is re-confirmed that the optimum river-flow has a dynamic role in pollution remediation especially for HMs in the polluted urban rivers.

Policy Implications and Future Research Perspective

Industries of urban areas altogether consume significant amounts of groundwater and create large volumes of polluted discharge. Thus, urban areas are suffering a shortage of sustainable water quality. As an active solution, government organizations can take necessary steps to maintain optimum recommended water-flow in the urban rivers as a part of the Delta Plan, 2100 (BDP 2100), announced by Bangladesh Government.

i. To ensure the minimum required water-flow in the urban river, a policy should be ensured for continuous 49.87 m³/sec water-flow from up-stream for sustainable water quality throughout the year.

Matal	Flow co-efficient*									
Ivietai	P ₁	P_2	P_{3}	P_4	P_{5}					
Cr	0.2827	-0.7139	0.6245	0.0516	-0.102					
Cd	0.2827	-0.1166	0.3163	-0.2226	0.1449					
Pb	0.2827	-0.8739	0.7266	0.04106	-0.1207					
Cu	0.2827	0.5948	0.0044	-0.0577	0.0309					
Mn	0.2827	-0.9271	0.7583	-0.1025	0.305					
Fe	0.2827	-1.357	1.889	-0.4841	0.8169					
Zn	0.2827	-0.3767	0.3	-0.0375	0.2045					
Ni	0.2827	-0.6587	0.4855	-0.0289	0.1009					

Table 5. Flow-coefficients for different metals.

* with 95% confidence intervals

Table 6. The level of heavy metal concentration for the minimum required water-flow.

	Metal	Cr	Cd	Pb	Cu	Mn	Fe	Zn	Ni
	Bangladesh standard (ppm)	0.5	0.5	0.1	0.5	0.3	1	5	0.1
Experimental	Minimum Water –Flow (m ³ /sec)	33.13	27.64	20.10	-	33.13	49.87	-	33.82
Experimental	Concentration (ppm)	0.218	0.390	0.008	-	0.267	0.911	-	0.064
Ampledical	Minimum Water- flow (m ³ /sec)				49	.87			
Anaryticai	Concentration (ppm)	0.095	0.005	BDL	0.126	0.099	0.911	0.091	BDL

BDL-Below Detection Level

It should be included in the 'Five Years Development Plan of Bangladesh' as well. Additionally, no further encroachment of the land down the river should be ensured.

- ii. In the Water Act of Bangladesh, 2013, there is a provision for using groundwater for industrial and irrigation purposes. However, some stakeholders suggested that there should be bans on the extensive withdrawal of groundwater. As an alternative, river water can be utilized after maintaining an optimum flow of 49.87 m³/sec.
- iii. A sustainable institutional framework, mentioning the requirement of the optimum flow of 49.87 m³/sec, can be put in place for the Bangladesh Water Development Board (BWDB) to implement the Master Plan (BDP 2100) for sustainable water management.
- iv. Presently, Water Resources Planning Organization (WARPO) is officially responsible for implementing the Water Act of Bangladesh, 2013. WARPO may include the requirement of optimum water-flow in the water governance of Bangladesh.
- v. Urban rivers in Bangladesh directly or indirectly receive large amounts of untreated industrial waste and effluents from a range of industries. These actors are not well aware of water issues or their exposure to water hazards imposed by their industries. Therefore, after being informed, illegal industries that distort the natural channel should be removed immediately from the riverbank.

Conclusions

Increased HMs concentration and loss of water-flow can have a detrimental effect on the ecological health of any urban river and its surrounding area. Compared to the other rivers in Bangladesh and around the world, the Turag River has been consistently exposed to a variety of HMs contamination. Thus, this river captured a more comprehensive scenario to investigate the correlation between HMs concentration and waterflow. The concentration of the selected heavy metals was evaluated in the water. Moreover, its current state has been compared to other countries and rivers around the world. The current state of the HMs concentration is frightfully exceeded the EQS for Bangladesh. The water-flow rate has also been observed to have reduced by more than 80% in the past two decades. Variation in the concentration of HMs was analyzed with the water-flow, distance from pollution sources, and different seasons (i.e., wet and dry). An empirical relation between HMs concentration and waterflow has been modeled and investigated to tackle the risk of HMs in river water. Thereby, a minimum waterflow requirement of 49.87 m³/sec has been analytically determined and verified to maintain the permissible level of HMs concentration for the ecological balance of the river.

Formulation of necessary policies and their strict enforcement is to be immediately ensured to increase the water-flow to minimize the perilous states of HMs concentration in the river water. The proposed modeling and analysis of the minimum water-flow requirement could also be studied for the other rivers as a pragmatic solution to tackle the risk of HMs and to save the ecological balance of the river and its surrounding area. Implementing of this method to the other polluted urban rivers of the world, optimum water-flow can be estimated as a part of their sustainable development program.

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Conflicts of Interest

The authors declare no conflict of interest.

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