

Original Research

# Startup and Performance Stability of a Nitrification-Anammox Reactor Using Granular Sludge

Bolin Li\*, Wenqin Zhang, Xueping Yan, Xin Huang, Jiangtao Li, Ye Li

School of Resources and Environmental Engineering, Wuhan University of Technology,  
Wuhan 430070, PR China

Received: 6 May 2016

Accepted: 6 August 2016

## Abstract

A nitrification-anammox single-stage autotrophic nitrogen-removal system was started and achieved stable, efficient operation. The startup process occurred in three stages: aerobic granulation, nitrification regulation, and autotrophic nitrogen removal. Granular sludge was successfully incubated after 57 d of acclimation. Through an operating strategy of gradual reduction in organic loading and an increase in ammonia nitrogen loading, stable nitrification was successfully achieved (average  $\text{NH}_4^+\text{-N}/\text{NO}_2^-$  ratio maintained at 1.32). Nitrification was coupled with anammox to achieve autotrophic nitrogen removal. The nitrogen loading rate was increased to  $0.3 \text{ kgN}/(\text{m}^3\cdot\text{d})$  and the removal rate reached  $0.26 \text{ kgN}/(\text{m}^3\cdot\text{d})$  after 50 d of operation. Test results showed that during the autotrophic nitrogen removal stage, anaerobic ammonia-oxidizing bacteria were considerably enriched, and that there was an organic balance between these and ammonia-oxidizing bacteria in the system.

**Keywords:** nitrogen removal, nitrification-anammox, stability, granular

## Introduction

The nitrification-anammox single-stage autotrophic nitrogen-removal process has the potential for broad application in the treatment of wastewater with a low carbon/nitrogen ratio [1-2]. In existing technologies, nitrification and its coupling with anammox are rate-limiting steps in the whole process and are critical breakthrough points for increasing the processing capacity [3-5].

The key to achieving an effective nitrification process is to enrich ammonia-oxidizing bacteria (AOB) while

inhibiting nitrite oxidizing bacteria (NOB). This enables AOB to have a competitive advantage and to achieve nitrite accumulation. Researchers have conducted numerous studies on ways to achieve nitrification, and have achieved nitrite accumulation by controlling one or more parameters, e.g., pH, dissolved oxygen (DO), free ammonia (FA), or free nitrous acid (FNA) [6-8]. However, NOB are highly adaptive and gradually resume their activity after the system has operated for a certain period. This makes it difficult to maintain long-term stability of the nitrification process.

DO is a critical factor used to achieve efficient and stable operation of the single-stage reactor. Low DO has a reversible inhibitory effect on anammox, whereas the inhibitory effect of high DO is irreversible [9-10]. Liu

\*e-mail: bolly1221@163.com

[11] and Vlaeminck et al. [12] controlled DO within a low range (0.2-1.0 mg/L) to facilitate the anammox process. However, it is very difficult to harmonize the contradictory DO demands of AOB and anaerobic ammonia-oxidizing bacteria (AAOB). Stable nitrite accumulation is essential to the operation of the autotrophic nitrogen removal system. In view of the aforesaid contradictory DO demands, Persson [8] and Gilbert et al. [13] achieved sludge granulation to enable a gradient distribution of DO in the granules. This led to the formation of aerobic and anaerobic microenvironments on the surface and in the interior, respectively, providing suitable dissolved oxygen conditions for symplastic growth of AOB and AAOB.

Tests proved that a gradient distribution of DO in granules simultaneously satisfied the antagonistic DO demands of AOB and AAOB while maintaining large amounts of biomass. However, because AAOB grows at an especially slow rate, adjusting the control parameters usually has an impact on the quantities and structures of microflora in the system. This can disrupt the microflora balance within the granular sludge, which is more difficult to restore than in flocculent sludge [14]. Consequently, the autotrophic nitrogen removal system would fail to maintain efficient and stable performance of nitrogen removal. Therefore, stricter requirements must be imposed on control of the operating parameters, such as loading, pH, and DO. To date, there are no effective control strategies for such parameters.

This study focuses on a stable startup process and operating strategy of a single-stage autotrophic nitrogen-removal system. Regular flocculent-activated sludge was inoculated and rapidly granulated in an SBR reactor under non-real-time control conditions. Based on the characteristics of ammonia nitrogen consumption and nitrite nitrogen accumulation, a control method for parameters (DO and pH) was established to achieve stable nitrification. Nitrification was coupled with the anammox process, and a control strategy for the single-stage autotrophic nitrogen removal system was proposed according to the differences in physiological properties among AOB, AAOB, and NOB. Based on this control strategy, stable operation of the single-stage autotrophic nitrogen-removal system was achieved, and the advantages of granule-enriched anammox biomass for the nitrogen-removal system will be discussed.

## Material and Methods

### Reactor and Operational Strategy

The SBR reactor was made of acrylic resin and had a working volume of 2.5 L, and a height/diameter ratio of 2.7. The exchange ratio ( $V_{\text{Effluent}}/V_{\text{Reactor}}$ ) was fixed and maintained at 0.5. During the experiment, the reaction temperature was kept near 25-30°C. The aeration rate was controlled by a gas rotameter. An overhead mechanical stirrer was used in the reactor at a fixed speed of 100 r/min.

The SBR was inoculated with 1.5 L of activated sludge collected from the Tangxunhu domestic sewage treatment plant in Wuhan, Hubei. The mixed liquid suspended solids (MLSS) value of the seeded sludge was 8-9 g/L.

Synthetic wastewater was used as influent in this study. The pH was controlled (7.8-8.2) by dosing with  $\text{NaHCO}_3$ . The synthetic wastewater was supplemented with mineral medium. We used synthetic medium containing COD (300 mg/L),  $\text{NH}_4^+\text{-N}$  (20 mg/L), TP (5 mg/L),  $\text{KH}_2\text{PO}_4$  (0.025 g/L),  $\text{CaCl}_2\cdot 2\text{H}_2\text{O}$  (0.3 g/L),  $\text{MgSO}_4$  (0.2 g/L), and  $\text{FeSO}_4$  (0.0065 g/L), and 1 mL/L trace element solution (as described in Wei et al., 2015) [15]. The  $\text{NH}_4^+\text{-N}$  concentration was increased to 200 mg/L to develop the nitrification process, and COD gradually was reduced to zero during this process.

### Analysis Methods

The pH, DO, and ORP parameters were measured with a portable and recordable probe (HACH, HQ40d). Measurements of ammonium, nitrite, nitrate, SS, and VSS were performed using standard methods (APHA 2005) [16]. The particle size distribution in random samples of granular sludge from the reactor was analyzed using a laser particle-size analyzer (Mastersizer 2000, Melvin British Company). Color photomicrographs (NIKON Mia2000 E200-F, Japan) were used for observation of sludge morphology.

### Quantitative PCR

A quantitative PCR system (Agilent Technologies, USA) was used to determine the distributions (in triplicate) of microbes in the SBR reactor. The gene sequences of AOB, NOB, AAOB, and DNF rRNA were tested as described by Persson et al. [8].

## Results and Discussion

### Aerobic Granulation Process

#### *Strategies for Sludge Granulation*

We found that when the organic loading was high, partial filamentous bacteria in the sludge served as the framework for granulation, thereby accelerating the formation of granules [17]. In this study, within 17 d after startup of the SBR system, organic loading was maintained at 2.0 kg COD/( $\text{m}^3\cdot\text{d}$ ), which facilitated the formation of microbial aggregates, and COD removal efficiency was maintained at 90% or above (Fig. 1). On Day 17 when the sludge expanded due to aeration failure, the organic loading was down-regulated to 1.2 kg COD/( $\text{m}^3\cdot\text{d}$ ), after which point the system gradually resumed. Organic loading was maintained at this level during the subsequent incubation process.

The sedimentation time for the sludge in the reactor was gradually reduced from an initial 20 min to 5 min.

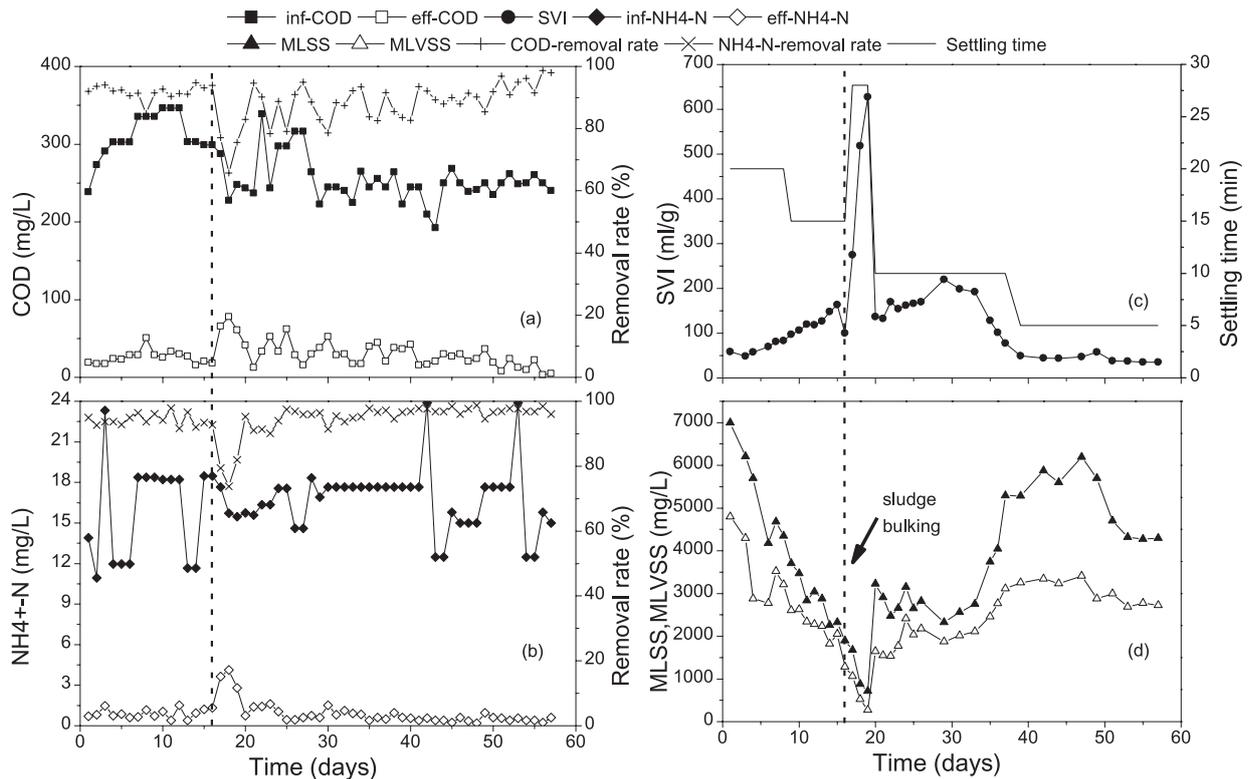


Fig. 1. Performance of the SBR system in the aerobic granulation stage.

The hydraulic screening action not only eliminated the dispersed, flocculent sludge from the reaction system, improved the systemic sedimentation properties (Fig. 1), and promoted the formation of the granular sludge, but it also stimulated changes in cell surface properties that facilitated the formation of biological flocs [18], thereby further promoting sludge granulation.

According to the study results of Zheng and Kenji Furukawa et al. [19-20], a shorter reaction cycle could lead to more frequent discharge of cells. Moreover, the hydraulic screening action enabled microorganisms with excellent sedimentation properties to remain in the system, and created suitable growth conditions for sludge granules. Within 1-10 d, the hydraulic retention time (HRT) was set at 8 h. Excessive aeration resulted in serious degradation of organic components into inorganic compounds, so the HRT was adjusted to 6 h, after which the MLSS in the system sustained a steady increase.

As shown in Fig. 1, during days 1-17 the system operated continuously with stable COD and ammonia-nitrogen-removal efficiency, both of which were maintained at 90% or above. During the sludge expansion stage, when the removal efficiencies started dropping, the aeration rate was adjusted to 0.6 L/min, DO was maintained at 1.5 mg/L, the effluent exchange ratio was reduced, and fresh sludge was replenished expediently. Afterward, the SVI of the system was maintained at 170 mL/g or below, sludge expansion was obviously alleviated, and MLSS quickly resumed a normal level of 2,600-3,200 mg/L. The reactor resumed stable removal efficiency on Day 28.

### Sludge Morphology

The SBR was operated for nearly two months under aerobic conditions for sludge granulation. The MLSS of the seeding sludge decreased from 8,000 mg/L to 6,980 mg/L after 7 d acclimation. Some tiny dark brown granules could be observed during stable operation. After sludge expansion on Day 17, the MLSS abruptly declined to 1,900 mg/L and the sludge appeared loose. On Day 28, the system gradually recovered and the sludge resumed its previous morphology (grayish black, small granules). On Day 57 the sludge had a more compact, dense structure with distinct granular boundaries (Fig. 2).

At the initial stage, the average particle size of the system was 97.478  $\mu\text{m}$ . After 57 d of operation, the average particle size increased to 308.214  $\mu\text{m}$ . It is widely accepted that if the proportion of granules >200  $\mu\text{m}$  is more than 10%, then the sludge has been successfully granulated [21]. On Day 49, more than 69% of the sludge granules in the system were greater than 200  $\mu\text{m}$ , indicating successful aerobic sludge granulation in the reaction system.

### The Achievement of Partial Nitrification

The saturation constant for oxygen is 0.2-0.4 mg/L for AOB and 1.2-1.5 mg/L for NOB [22]. This indicates that under low DO, AOB grew faster than NOB, so AOB could be enriched by the optimization of operating conditions. To investigate the feasibility of achieving stable nitritation in the SBR through DO control and the stability of the nitritation process, from Day 62 to 83 (Stage I) the aeration

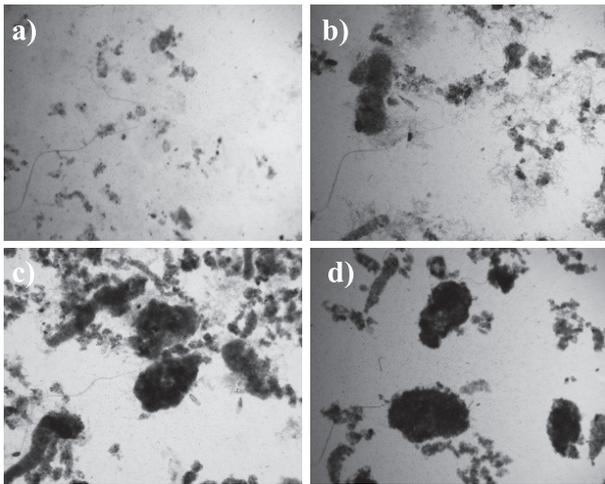


Fig. 2. Optical microscope photographs (40 $\times$ ): a) 1d after seeding, b) 10d after seeding, c) 35d after seeding, d) 56d after seeding.

rate was maintained at  $0.4 \pm 0.1$  L/min. Under the fixed oxygen supply mode, COD gradually decreased from 220 to 110 mg/L and the reactor exhibited excellent nitrification performance, so the influent  $\text{NH}_4^+\text{-N}$  concentration was increased gradually from 30 to 70 mg/L (Fig. 3).

From Day 84-123 (Stage II), the influent  $\text{NH}_4^+\text{-N}$  concentration was increased to 100 mg/L, while COD decreased to zero. At a fixed aeration rate, as  $\text{NH}_4^+\text{-N}$  was gradually consumed, the nitrification reaction slowed down due to reduced substrate concentration. As the DO level required for nitrification was lower than that for nitritation, the DO level in the system quickly increased and the nitrogen removal rate of the reactor declined. Thereafter, nitrite oxidation occurred, leading to reduced concentration of accumulated  $\text{NO}_2^-\text{-N}$  and to the reduction of DO to the “characteristic point” [23] (Fig. 4), which could be used as the control point for the nitritation process.

Thereafter, the influent  $\text{NH}_4^+\text{-N}$  concentration was increased to 200 mg/L. To maintain the nitritation

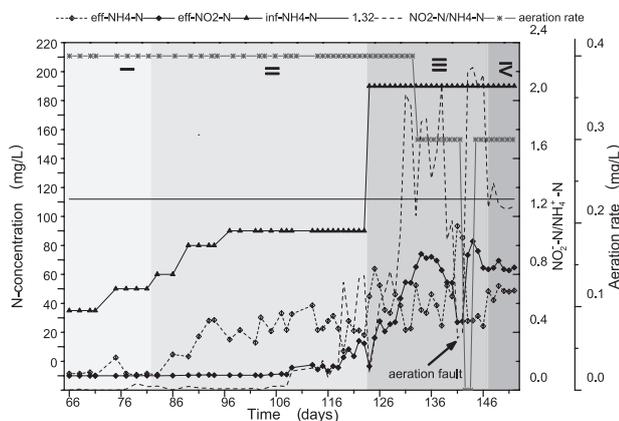


Fig. 3. Variations in influent/effluent concentrations of  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_2^-\text{-N}$ ,  $\text{NO}_2^-\text{-N}/\text{NH}_4^+\text{-N}$  and aeration rate in the nitritation process.

reaction, during stage III (Day 123-148) the aeration rate was reduced from 0.4 to 0.3 L/min, and nitrite nitrogen started accumulating rapidly in the system, reaching 86.2 mg/L. During Day 142-143, due to aeration failure, the aeration rate of the system was reduced to zero, nitritation was seriously inhibited, and the  $\text{NO}_2^-\text{-N}$  concentration dropped sharply. After troubleshooting the aeration failure, the aeration rate resumed at 0.3 L/min, and the  $\text{NO}_2^-\text{-N}$  accumulation rate increased and stabilized at 50%. After 148 days, the  $\text{NO}_2^-\text{-N}/\text{NH}_4^+\text{-N}$  ratio of the system stabilized at 1.32 (Stage IV), providing the foundation for subsequent coupling with anammox.

As shown in Fig. 3, during the whole nitritation stage, the amount of accumulated  $\text{NO}_2^-\text{-N}$  slowly increased as nitrogen loading rate (NLR) increased. During the late phase, due to denitrification in the reactor, the amount of accumulated  $\text{NO}_2^-\text{-N}$  declined. Afterward, as NLR continued to increase, the aeration rate and aeration time were adjusted and the AOB became the dominant microflora. The conversion rate of  $\text{NH}_4^+\text{-N}$  to  $\text{NO}_2^-\text{-N}$  reached 95% on Day 142. This finding is in full agreement with the results of a study of single-stage nitrogen removal in granular sludge, where NOB repression was also related to the residual ammonium concentration [14].

In addition, alkaline conditions facilitated the formation of FA, inhibited NOB activity by reducing the needed substrate, and provided a substrate for AOB [24]. The particular pH conditions also imposed a considerable direct effect on microorganisms. Optimal growth occurred at pH 7.0-8.5 for AOB and at pH 6.0-7.5 for NOB. At high pH, AOB had greater activity than did NOB, leading to selection for nitritation reactions rather than nitrification reactions. In contrast, the nitritation reaction system could be destroyed by low pH. Figure 4 shows that if alkalinity were insufficient, the  $\text{NH}_4^+\text{-N}$  was almost completely oxidized at 135 min, and the pH value started to increase afterward.

In consideration of the corresponding DO changes, pH and DO were used as parameters to control the reaction time, and the nitritation process was controlled by changes in initial alkalinity and DO level during the reaction. Therefore, adequate alkalinity and a rational oxygen supply mode did not inhibit the nitritation reaction rate, and facilitated the establishment of a stable control process for nitritation.

## Denitrification Performance of the CANON System

### Start-up and Stable Operation

During the startup stage of the nitritation process, the activity of NOB was inhibited and AOB continuously grew and reproduced. This was accomplished by enabling AOB and NOB to have different properties by manipulating various factors, such as DO, pH, COD, and FA. During the initial stage of the autotrophic nitrogen-removal process there was a small number of AAOBs in the system. Because the doubling time of AOB is more than 10 times

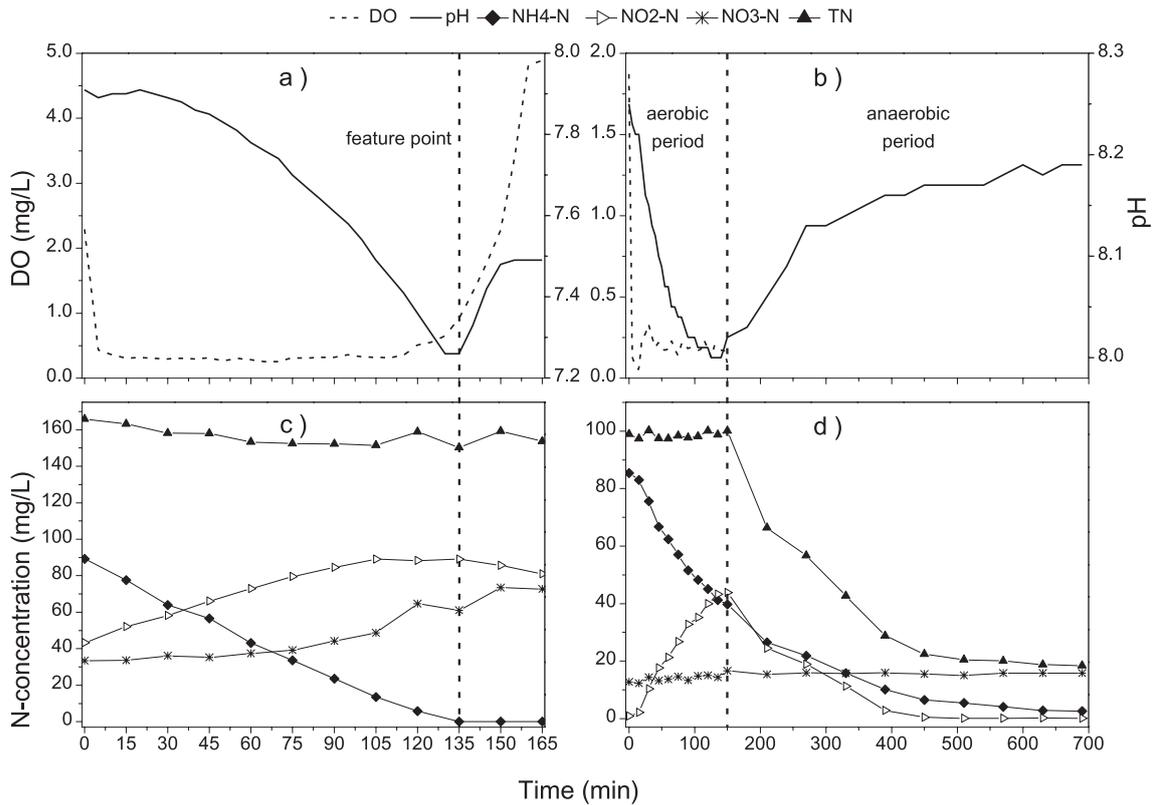


Fig. 4. Time course of parameters and substrates measured in two experimental phases: (a-c) nitrification regulation and (b-d) autotrophic nitrogen removal.

that of AAOB, the HRT was extended from 6 to 24 h, and 4 mg/L  $N_2H_4$  was added as an intermediate product for stimulating the AAOB growth. These actions led to enhanced anammox [25] and rapid AAOB enrichment.

During Days 161-218 (Stage II), the nitrogen removal efficiency of the system gradually increased to 75.8% as a consequence of AAOB enrichment. The aeration time was extended from 130 to 145 min on Day 220 to ensure that  $NO_2^-$ -N formation demand was met for the anammox

process. Accordingly, the effluent  $NH_4^+$ -N concentration decreased rapidly, which allowed the nitrogen removal efficiency and the nitrogen removal rate (NRR) to increase to 79.5% and 0.18 kgN/(m<sup>3</sup>·d), respectively (Fig. 5).

To increase the hydraulic loading of the system, during Stage III the HRT was shortened to 12 h and the anaerobic reaction time was shortened. After this, the conversion rate of  $NH_4^+$ -N and total nitrogen removal efficiency decreased. Thereafter,  $NH_4^+$ -N gradually accumulated in the reactor, AAOB activity declined, and the total nitrogen removal rate fell to 37.2%. When the aeration time was increased to 150 min, the system gradually resumed its nitrogen removal performance. The nitrogen removal loading and total nitrogen removal rate were stabilized at 0.26 kgN/(m<sup>3</sup>·d) and 84.3%, respectively; higher than the total nitrogen removal rate of 74.4±7.1% and 72±9% obtained by C.M. Castro-Barros et al. [1] and Peili Lu et al. [26] with different types of reactors and conditions.

#### Performance of a Typical Cycle

Different reactors have different DO demands for startup and operation of the CANON process. The DO level in an SBR reactor is normally maintained at 0.5-0.8 mg/L. Helmer-Madhok et al. [27] found that when DO was 0-0.7 mg/L, DO was positively correlated with the autotrophic nitrogen removal efficiency, and a further increase in DO level would destroy the system. Li et al.

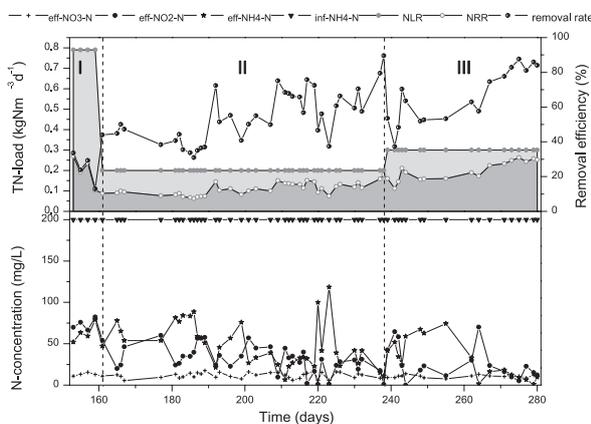


Fig. 5. Variation of nitrogen loading rate (NLR), nitrogen removal rate (NRR), removal efficiency, and N concentration of the autotrophic nitrogen removal process.

[28] successfully cultured CANON granular sludge in the SBR reactor at a DO level of 0.3-0.5 mg/L, and the TN removal efficiency reached 63.7%. Figure 4 shows that the DO concentration ranged from 0.09 to 0.39 mg/L during the CANON process, with a mean value of 0.22 mg/L. These are similar to the results for integrated anammox processes reported abroad.

Both AOB and AAOB required suitable alkalinity, and the distribution ratios of reaction substrates  $\text{NH}_3$  and  $\text{HNO}_2$  in the liquid phase were also influenced by pH. Therefore, it is necessary to adjust the pH carefully to enhance anammox without compromising ammonium oxidation and thereby increase nitrogen removal performance. Based on previous studies of the effects of pH on the CANON process, it is preferable to keep pH at 8 [9, 29-30].

Fig. 4 shows the conversion status of substrates as well as changes in the pH and DO in the reaction system within a single cycle. The first 1.5 h of the cycle was the aerobic stage. At the end of aeration, the  $\text{NO}_2^- \text{-N}/\text{NH}_4^+ \text{-N}$  ratio was 1.11, and  $\text{NO}_3^- \text{-N}$  and TN concentrations remained almost unchanged. During the anaerobic stage, pH gradually increased to an extent that offset the amount of alkali consumed in the nitrification stage. This enabled the system to achieve partial acid-alkaline balance, and also reflected the resource-saving nature of the CANON process.

### Characterization of the Microbial Community

During the late phase of acclimation, the absolute abundance of AOB was 7.4% at the nitrification stage (Fig. 6), indicating considerable enrichment of AOB compared with that in the initial seeding sludge (absolute abundance 0.87%). The absolute abundance slightly decreased during the autotrophic nitrogen removal stage. The NOB had a high concentration in the seeding sludge, while it sharply decreased during the nitrification stage and further declined to its lowest level in the CANON stage. The DNF exhibited the same patterns of change as the NOB. The absolute

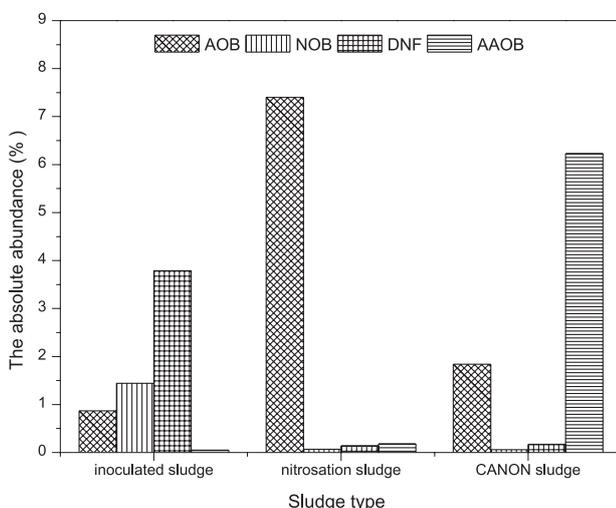


Fig. 6. Absolute abundance of functional bacteria in different stages.

abundance of AAOB increased by 135-fold during the CANON stage, compared with that in the seeding sludge. The NOB was not completely eliminated, and the microflora in the system exhibited obvious structural changes. According to the results from previous domestic and international PCR studies [13], it can be seen that the AAOB were successfully enriched, and that AOB and AAOB became the predominant bacteria in the system.

When AOB and AAOB coexist in the system, AOB provides AAOB with reaction substrate and AAOB eliminates the substrate inhibition for AOB. However, they do compete for the same substrate when FA is insufficient in the system [31]. AOB is an aerobe while AAOB is an obligate anaerobe, the growth and metabolic processes of which are inhibited in the aerobic environment. Thus the relationship between the two bacteria is complicated. In a CANON system with adequate sludge granulation, the DO concentration gradient inside the granular sludge enabled the granular sludge to form a layered structure with aerobic exterior and anaerobic interior environments. This oxidative structure allowed AOB and NOB to coexist harmoniously in the same system.

### Conclusions

Rapid startup of a nitrification-anammox process was achieved in a lab-scale SBR reactor with granular sludge. In the presence of ammonium, dissolved oxygen was an effective control parameter for suppressing the undesired NOB activity. AAOB were shown to grow in the CANON system and an organic balance between AAOB and AOB was achieved.

1. Nitrified sludge from a domestic wastewater treatment plant was inoculated and successfully granulated after 57 days of incubation. The NLR of the system was 0.08 kgN/(m<sup>3</sup>·d), and nitrogen removal efficiency reached 90% or above. The sludge gradually changed to dark brown and more than 50% of the granules were larger than 1 mm.
2. Partial nitrification was stably maintained long-term in a granular reactor with an average  $\text{NH}_4^+ \text{-N}/\text{NO}_2^- \text{-N}$  ratio of 1.32, when the AOB population had a higher affinity for DO than that of the NOB. The microbial structure and  $\text{NO}_2^- \text{-N}$  accumulation rates were controlled using a strategy based on influent loads, aeration rate, and alkalinity.
3. Rapid AAOB enrichment and organic balance between AAOB and AOB can be achieved in a granular reactor under oxygen-limiting conditions. The nitrogen removal loading and total nitrogen removal rate were about 0.26 kgN/(m<sup>3</sup>·d) and 84.3%, respectively.

### Acknowledgements

This research is supported by the Natural Science Fund of Hubei Province (grant No. 2014CFB285),

the Wuhan Science and Technology Foundation (No. 2014060101010065) and the Wuhan Environmental Protection Scientific Research Projects (No. 20151h0035).

### References

- CASTRO-BARROS, C.M., DAELMAN, M.R.J., MAMPAEY, K.E. Effect of aeration regime on  $N_2O$  emission from partial nitrification-anammox in a full-scale granular sludge reactor. *Water Research* **68** (1), 793, **2015**.
- DAVEREY A., SU S.H., HUANG Y.T., CHEN S.S., SUNG S.W., LIN J.G. Partial nitrification and anammox process: A method for high strength optoelectronic industrial wastewater treatment. *Water Research* **47** (9), 2919, **2013**.
- LACKNER S., GILBERT E.M., AGRAWAL S., KARST S.M., HORN H., NIELSEN P.H. Low Temperature Partial Nitrification/Anammox in a Moving Bed Biofilm Reactor Treating Low Strength Wastewater. *Environmental Science Technology* **48**, 8784, **2014**.
- LIN J.G., DAVEREY A., CHEN Y.C., DUTTA K., HUANG Y.T. Start-up of simultaneous partial nitrification, anammox and denitrification (SNAD) process in sequencing batch biofilm reactor using novel biomass carriers. *Bioresource Technology* **190**, 480, **2015**.
- CHAI L.Y., ALI M., MIN X.B., SONG Y.X., TANG C.J., WANG H.Y., YU C., YANG Z.H. Partial nitrification in an air-lift reactor with long-term feeding of increasing ammonium concentrations. *Bioresource Technology* **185**, 134, **2015**.
- HISASHI S., RATHNAYAKE M.L.D., RATHNAYAKE M.O., ISHII S., SEGAWA, T., OKEBE S. Effects of dissolved oxygen and pH on nitrous oxide production rates in autotrophic partial nitrification granules. *Bioresource Technology* **197**, 15, **2015**.
- ZHANG X.J., LI D., LIANG Y.H., ZENG H.P., HE Y.P., FAN D., ZHANG J. Start-up, influence factors, and the microbial characteristics of partial nitrification in membrane bioreactor. *Desalination and Water Treatment* **54** (3), 581, **2015**.
- PERSSON F., SULTANAR., SUAREZ M., HERMANSSON M., PLAZA E., WILEN B.M. Structure and composition of biofilm communities in a moving bed biofilm reactor for nitrification-anammox at low temperatures. *Bioresource Technology* **154**, 264- **2014**.
- VAN DER MEER J.R., ZEHNDER A.J.B. Enrichment and characterization of an anammox bacterium from a rotating biological contactor treating ammonium-rich leachate. *Archives of Microbiology* **175** (3), 198, **2015**.
- LOTTI T., KLEEREBEZEM R., HU Z., KARTAL B., JETTEN M.S.M., VAN LOOSDRECHT M.S.M. Simultaneous partial nitrification and anammox at low temperature with granular sludge. *Water Research* **66** (3), 111, **2014**.
- LIU T., LI D., ZENG H.P., CHANG X.Y., ZHANG J. Microbial characteristics of a CANON reactor during the start-up period seeding conventional activated sludge. *Water Science and Technology* **67** (3), 635, **2013**.
- VLAEMINCK S.E., TERADA A., SMETS B.F., VAN DER LINDEN D., BOON N., VERSTRAETE W., CARBALLA M. Nitrogen removal from digested black water by one-stage partial nitrification and anammox. *Environment Science and Technology* **43**, 5035, **2009**.
- GILBERT E.M., AGRAWAL S., KARST S.M., HORN H., NIELSEN P.H., LACKNER S. Low temperature partial Nitrification/Anammox in a moving bed biofilm reactor treating low strength wastewater. *Environmental Science & Technology* **48**, 8784, **2014**.
- FERNÁNDEZ I., VÁZQUEZ-PADÍN J.R., MOSQUERA-CORRAL A., CAMPOS J.L., MÉNDEZ R. Biofilm and granular systems to improve ANAMMOX biomass retention. *Biochemical Engineering Journal* **42** (3), 308, **2008**.
- WEI D., DU B., ZHANG J., HU Z., LIANG S., LI Y.R. Composition of extracellular polymeric substances in a partial nitrification reactor treating high ammonia wastewater and nitrous oxide emission. *Bioresource Technology* **190**, 474, **2015**.
- APHA. Standard Methods for the Examination for Water and Wastewater, 21<sup>th</sup> Ed. American Public Health Association, Washington, DC., USA, **2005**.
- PENG Y.Z., WANG S.Y., MA B., WANG S.Y., ZHU G.B. Anaerobic ammonium oxidation in traditional municipal wastewater treatment plants with low-strength ammonium loading: Widespread but overlooked. *Water Research* **84**, 66, **2015**.
- VÁZQUEZ-PADÍN J.R., MOSQUERA-CORRAL J.A., CAMPOS J.L., REVSBECH N.R. Microbial community distribution and activity dynamics of granular biomass in a CANON reactor. *Water Research* **44** (15), 4359, **2010**.
- ZHENG P., WANG L., XING Y.J., WEI L., YANG J., ABBAS G., LIU S., HE Z.F., ZHANG J.Q., ZHANG H.T., LU, H.F. Effect of particle size on the performance of autotrophic nitrogen removal in the granular sludge bed reactor and microbiological mechanisms. *Bioresource Technology* **157**, 240, **2014**.
- FURUKAWA K., ZHANG L., YANG J.C. Stable and high-rate nitrogen removal from reject water by partial nitrification and subsequent anammox. *Bioscience and Bioengineering* **110** (4), 441, **2010**.
- PIJUAN M., WERNER U., YUAN Z.G. Reducing the startup time of aerobic granular sludge reactors through seeding floccular sludge with crushed aerobic granular aerobic granules. *Water Research* **45** (16), 5075, **2011**.
- ZHENG P., ABBAS G., WANG L., ZHANG J.P. Oxygen Transfer Characteristics in a Pilot-Scale Airlift Internal-Loop Bioreactor for Simultaneous Partial Nitrification and Anaerobic Ammonia Oxidation. *Environmental Engineering Science* **31** (8), 453, **2014**.
- LI D., LIANG Y.H., ZENG H.P., ZHANG C.D., ZHANG J. Rapid start-up and microbial characteristics of partial nitrification granular sludge treating domestic sewage at room temperature. *Bioresource Technology* **196**, 741, **2015**.
- DURAN U., DEL RIO A.V., CAMPOS J.L., MOSQUERA-CORRAL A., MENDEZ R. Enhanced ammonia removal at room temperature by pH controlled partial nitrification and subsequent anaerobic ammonium oxidation. *Environmental Technology* **35** (4), 383, **2014**.
- YAO Z.B., CAI Q., ZHANG D.J., XIAO P.Y., LU P.L. The enhancement of completely autotrophic nitrogen removal over nitrite (CANON) by  $N_2H_4$  addition. *Bioresource Technology* **146**, 591, **2013**.
- LU P.L., XIAO P.Y., ZHANG D.J. Effect of trace hydrazine addition on the functional bacterial community of a sequencing batch reactor performing completely autotrophic nitrogen removal over nitrite. *Bioresource Technology* **175**, 216, **2015**.
- HELMER M.C., SCHMID M., FILIPOV E., GAUL T., HIPPEN A., ROSENWINKEL K.H., SEYFRIED C.F., WAGNER M., KUNST S. Deammonification in biofilm systems: population structure and function. *Water Science and Technology* **46** (1-2), 223, **2001**.

28. LI D., LIANG Y.H., ZHANG X.J., ZENG H.P., YANG Z., CUI S.M., ZHANG J. Stability and nitrite-oxidizing bacteria community structure in different high-rate CANON reactors. *Bioresource Technology* **175**, 189, **2015**.
29. LI K., FANG F., GUO J.S. Performance of one-stage autotrophic nitrogen removal in a biofilm reactor with low C/N ratio. *Environmental Technology* **36** (14), 1819, **2015**.
30. XING B.S., JI Y.X., YANG G.F., CHEN H., NI W.M., JIN R.C. Start-up and stable operation of partial nitrification prior to ANAMMOX in an internal-loop airlift reactor. *Separation and Purification Technology* **120**, 458, **2013**.
31. LACKNER S., THOMA K., GILBERT E.M., GANDER W., SCHREFF D., HORN H. Start-up of a full-scale deammonification SBR-treating effluent from digested sludge dewatering. *Water Science and Technology* **71** (4), 553, **2015**.